

Chapter 2 Properties of Matter

Section 2.3 Chemical Properties

(pages 54–58)

*This section discusses chemical properties and describes clues that may show that a chemical change has taken place.***Reading Strategy** (page 54)

Relating Text and Visuals. As you read, complete the table by finding examples of the clues for recognizing chemical changes in Figures 19 and 20. For more information on this Reading Strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.

Recognizing Chemical Changes	
Clue	Example
Change in color	Copper nail changing color from red to green when exposed to moist air
Production of gas	Formation of carbon dioxide gas when vinegar is added to baking soda
Formation of precipitate	Formation of cottage cheese curds when acid is added to milk.

Observing Chemical Properties (pages 54–55)

1. Is the following sentence true or false? The substances in paraffin do not change when a candle burns. false
2. Circle the letters of the compounds formed when a candle burns.
 - a. paraffin
 - b. hydrogen
 - c. water
 - d. carbon
3. What is a chemical property? A chemical property is any property that produces a change in the composition of matter.
4. Is the following sentence true or false? Flammability is a material's ability to burn in the presence of carbon dioxide. false
5. The property that describes how readily a substance combines chemically with other substances is reactivity.
6. Circle the letter of each property that is a chemical property.

a. hardness	b. density
<input checked="" type="radio"/> c. flammability	<input checked="" type="radio"/> d. reactivity
7. Is the following sentence true or false? Nitrogen is a more reactive element than oxygen. false

23 Chemical Properties Answer Key

IM Harris

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Decoding **23 Chemical Properties Answer Key**: Revealing the Captivating Potential of Verbal Expression

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