

Chemthink: CHEMICAL REACTIONS

- a) Go to www.chemthink.com. Log in using your user name and password. (If you forgot, you need to WAIT QUIETLY until your teacher is done introducing the lesson.)
- b) Scroll to **CHEMICAL REACTIONS** then the tutorial under the Chemical Reactions menu.
- c) Complete the **tutorial**. As you complete the tutorial, write down the answers to the questions below.
- d) Once you have successfully completed the tutorial and this worksheet, complete the **Question Set**.

CHEMICAL REACTIONS TUTORIAL QUESTIONS:

- Starting materials in a chemical reaction are called _____.
- The ending materials in a chemical reaction are called _____.
- The arrow indicates a _____ has taken place.
- All reactions have one thing in common: there is a _____ of chemical bonds.
- Chemical reactions always involve _____, old bonds, _____, new bonds, or both.
- In all reactions we still have all of the _____ at the end that we had at the start.
- In every reaction there can never be any _____ atoms or _____ in the atoms that are already there.
- Chemical reactions only _____ in the atoms that are already there.
- Let's represent a reaction on paper. For example, hydrogen gas (H_2) reacts with oxygen gas (O_2) to form water (H_2O):



If we use only the atoms shown, we'd have _____ atoms of H and _____ atoms of O as reactants. This would make _____ molecule of H_2O , but we'd have _____ atom of O leftover. However, this reaction only makes H_2O .

Remember: reactions are not limited to 1 molecule each of reactants. We can use as many as we need to balance the chemical equation.

A balanced chemical reaction shows:

- a) What atoms are present before (in the reactants) and after (in the products)
 - b) **How many** of each reactant and product is present before and after.
10. So to make H_2O from oxygen gas and hydrogen gas, the balanced equation would be:



Which is the same as:

# of atoms in Reactants	Element	# of atoms in Products
	H	
	O	

- This idea is called the _____.
- There must be the same _____ and the same number of _____ before the reaction (in the reactants) and after the reaction (in the products).
- What is the balanced equation for this reaction? _____ Cu + _____ $O_2 \rightarrow$ _____ CuO

Answers To Chemthink Questions Tutorial Test

H. Stephen Stoker, Danny V. White



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