

# Periodic Table of the Elements

| 2<br>IA<br>2A                    | 13<br>IIIA<br>3A                   | 14<br>IVA<br>4A                 | 15<br>VA<br>5A                       | 16<br>VIA<br>6A                   | 17<br>VIIA<br>7A                     |
|----------------------------------|------------------------------------|---------------------------------|--------------------------------------|-----------------------------------|--------------------------------------|
| <b>Be</b><br>Beryllium<br>9.012  | <b>B</b><br>Boron<br>10.811        | <b>C</b><br>Carbon<br>12.011    | <b>N</b><br>Nitrogen<br>14.007       | <b>O</b><br>Oxygen<br>15.999      | <b>F</b><br>Fluorine<br>18.998       |
| <b>Mg</b><br>Magnesium<br>24.305 | <b>Al</b><br>Aluminum<br>26.982    | <b>Si</b><br>Silicon<br>28.086  | <b>P</b><br>Phosphorus<br>30.974     | <b>S</b><br>Sulfur<br>32.065      | <b>Cl</b><br>Chlorine<br>35.453      |
| <b>Ca</b><br>Calcium<br>40.078   | <b>Ga</b><br>Gallium<br>69.723     | <b>Ge</b><br>Germanium<br>72.61 | <b>As</b><br>Arsenic<br>74.922       | <b>Se</b><br>Selenium<br>78.96    | <b>Br</b><br>Bromine<br>79.904       |
| <b>Sr</b><br>Strontium<br>87.62  | <b>In</b><br>Indium<br>114.818     | <b>Sn</b><br>Tin<br>118.71      | <b>Sb</b><br>Antimony<br>121.760     | <b>Te</b><br>Tellurium<br>127.6   | <b>I</b><br>Iodine<br>126.904        |
| <b>Ba</b><br>Barium<br>137.327   | <b>Tl</b><br>Thallium<br>204.383   | <b>Pb</b><br>Lead<br>207.2      | <b>Bi</b><br>Bismuth<br>208.980      | <b>Po</b><br>Polonium<br>[209]    | <b>At</b><br>Astatine<br>209 [97]    |
| <b>Ra</b><br>Radium<br>226.025   | <b>Uut</b><br>Ununtrium<br>unknown | <b>Fl</b><br>Flerovium<br>[289] | <b>Uup</b><br>Ununpentium<br>unknown | <b>Lv</b><br>Livermorium<br>[293] | <b>Uus</b><br>Ununseptium<br>unknown |

|                   |   |                                       |  |  |  |   |   |   |   |   |   |  |  |   |   |
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| Lanthanide Series | 57<br><b>La</b><br>Lanthanum<br>138.905 | 58<br><b>Ce</b><br>Cerium<br>140.115  | 59<br><b>Pr</b><br>Praseodymium<br>140.908 | 60<br><b>Nd</b><br>Neodymium<br>144.24 | 61<br><b>Pm</b><br>Promethium<br>144.913 | 62<br><b>Sm</b><br>Samarium<br>150.36   | 63<br><b>Eu</b><br>Europium<br>151.965  | 64<br><b>Gd</b><br>Gadolinium<br>157.25 | 65<br><b>Tb</b><br>Terbium<br>158.925   | 66<br><b>Dy</b><br>Dysprosium<br>162.50   | 67<br><b>Ho</b><br>Holmium<br>164.930   | 68<br><b>Er</b><br>Erbium<br>167.26    | 69<br><b>Tm</b><br>Thulium<br>168.934    | 70<br><b>Yb</b><br>Ytterbium<br>173.04  | 71<br><b>Lu</b><br>Lutetium<br>174.967  |
| Actinide Series   | 89<br><b>Ac</b><br>Actinium<br>227.028  | 90<br><b>Th</b><br>Thorium<br>232.038 | 91<br><b>Pa</b><br>Protactinium<br>231.036 | 92<br><b>U</b><br>Uranium<br>238.029   | 93<br><b>Np</b><br>Neptunium<br>237.048  | 94<br><b>Pu</b><br>Plutonium<br>244.064 | 95<br><b>Am</b><br>Americium<br>243.061 | 96<br><b>Cm</b><br>Curium<br>247.070    | 97<br><b>Bk</b><br>Berkelium<br>247.070 | 98<br><b>Cf</b><br>Californium<br>251.080 | 99<br><b>Es</b><br>Einsteinium<br>[254] | 100<br><b>Fm</b><br>Fermium<br>257.095 | 101<br><b>Md</b><br>Mendelevium<br>258.1 | 102<br><b>No</b><br>Nobelium<br>259.101 | 103<br><b>Lr</b><br>Lawrencium<br>[262] |

|              |                |                  |             |           |          |         |           |            |          |
|--------------|----------------|------------------|-------------|-----------|----------|---------|-----------|------------|----------|
| Alkali Metal | Alkaline Earth | Transition Metal | Basic Metal | Semimetal | Nonmetal | Halogen | Noble Gas | Lanthanide | Actinide |
|--------------|----------------|------------------|-------------|-----------|----------|---------|-----------|------------|----------|

## 5 2 Apply Answer Key Using The Periodic Table

**Rachel E. Green**



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