

## Section 5.3 Electron Configuration

In your textbook, read about ground-state electron configurations.

Use each of the terms below just once to complete the passage.

|                  |                           |                                     |             |
|------------------|---------------------------|-------------------------------------|-------------|
| Aufbau principle | electron configuration    | ground-state electron configuration | Hund's rule |
| lowest           | Pauli exclusion principle | spins                               | stable      |

The arrangement of electrons in an atom is called the atom's

(1) \_\_\_\_\_. Electrons in an atom tend to assume the arrangement that gives the atom the (2) \_\_\_\_\_ possible energy. This arrangement of electrons is the most (3) \_\_\_\_\_ arrangement and is called the atom's (4) \_\_\_\_\_.

Three rules define how electrons can be arranged in an atom's orbitals. The

(5) \_\_\_\_\_ states that each electron occupies the lowest energy orbital available. The (6) \_\_\_\_\_ states that a maximum of two electrons may occupy a single atomic orbital, but only if the electrons have opposite (7) \_\_\_\_\_. (8) \_\_\_\_\_ states that single electrons with the same spin must occupy each equal-energy orbital before additional electrons with opposite spins occupy the same orbitals.

Complete the following table.

| Element   | Atomic Number | Orbitals |    |                 |                 |                 | Electron Configuration |
|-----------|---------------|----------|----|-----------------|-----------------|-----------------|------------------------|
|           |               | 1s       | 2s | 2p <sub>x</sub> | 2p <sub>y</sub> | 2p <sub>z</sub> |                        |
| 9. Helium |               |          |    |                 |                 |                 | 1s <sup>2</sup>        |
| 10.       | 7             |          |    |                 |                 |                 |                        |
| 11. Neon  |               | ↑↓       | ↑↓ | ↑↓              | ↑↓              | ↑↓              |                        |

# 5 3 Distributing Electrons Study Guide Answer Key

**Clemens Wendtner**



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am writing to request a 15 day leave to be absent from my job duties due to medical reasons as my health seems to keep  
deteriorating my doctor advised me to take some time off work

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a simple guide october 5 2023 by tanu being unwell is a part of life and when it happens it s essential to inform your school  
about your illness and request sick leave

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school to be granted to name of the child form group first day of proposed absence last date of proposed absence expected  
date of return to school

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application can be submitted by the mother or father sick leave application for child respected principal my son is studying in



the class of play group in your school

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