

# Acid-Base Chemistry

## acids

- pH less than 7
- Turn litmus paper red
- Taste sour
- Feel irritating/corrosive
- Include fruit juices, soda, and coffee

## bases

- pH greater than 7
- Turn litmus paper blue
- Taste bitter or soapy
- Feel slippery
- Include baking soda, ammonia, and soap



neutralization  
reaction

**strong acid** + **strong base** → salt + water

example: **HCl** + **NaOH** → NaCl + H<sub>2</sub>O

## 20 Describing Acids And Bases Answers

**Ronald James Gillespie**



## 20 Describing Acids And Bases Answers:

Physical Chemistry and Its Biological Applications Wallace Brey, 2012-12-02 Physical Chemistry and Its Biological Applications presents the basic principles of physical chemistry and shows how the methods of physical chemistry are being applied to increase understanding of living systems Chapters 1 and 2 of the book discuss states of matter and solutions of nonelectrolytes Chapters 3 to 5 examine laws in thermodynamics and solutions of electrolytes Chapters 6 to 8 look at acid base equilibria and the link between electromagnetic radiation and the structure of atoms Chapters 9 to 11 cover different types of bonding the rates of chemical reactions and the process of adsorption Chapters 12 to 14 present molecular aggregates magnetic resonance spectroscopy and photochemistry and radiation This book is useful to biological scientists for self study and reference With modest additions of mathematical material by the teacher the book should also be suitable for a full year major s course in physical chemistry

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Science Examination Papers Including the Papers Set at the Evening Examinations ... ,1901      *Watts' Dictionary of Chemistry* Henry Watts, Henry Forster Morley, Matthew Moncrieff Pattison Muir, 1894      **The Journal of the Society of Chemical Industry** ,1892      *Journal of the Society of Chemical Industry* Society of Chemical Industry (Great Britain), 1892  
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