

20.1**DESCRIBING ACIDS AND BASES****SECTION REVIEW****Objectives**

- List the properties of acids and bases.
- Name an acid or base when given the formula.

Key Terms

- acid
- base

Part A Completion

After this completion exercise is done, your understanding of the concepts and terms is clear and reinforced in this section. Each blank has the compound with a name, about phrase, or a symbol.

Aqueous solutions of _____ 1 _____ are electrolytes. Acids react with many _____ 2 _____ to produce _____ 3 _____ gas. _____ 4 _____ are compounds that react with acids to form _____ 5 _____ and a salt.

Acids produce _____ 6 _____ ions when dissolved in water. There is a standard system for naming acids. For example, HBr is called _____ 7 _____ acid, while HNO_3 is called _____ 8 _____ acid.

Bases produce _____ 9 _____ ions when dissolved in water. Some compounds that are _____ 10 _____ are named in the same way as other ionic compounds.

1. _____

2. _____

3. _____

4. _____

5. _____

6. _____

7. _____

8. _____

9. _____

10. _____

Part B True-False

Circlely write *if these classes are true* (abbreviate *if*), *AC* (some classes *are*), *SC* (no class *are*), *NT*.

- _____ 11. A compound that contains hydrogen atoms will be an acid when dissolved in water.
- _____ 12. An acid contains one or more hydroxide ions.
- _____ 13. Chemists have a special system for naming bases.

20 Describing Acids And Bases Answer Key

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Ruch**



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Physical Chemistry and Its Biological Applications Wallace Brey, 2012-12-02 Physical Chemistry and Its Biological Applications presents the basic principles of physical chemistry and shows how the methods of physical chemistry are being applied to increase understanding of living systems Chapters 1 and 2 of the book discuss states of matter and solutions of nonelectrolytes Chapters 3 to 5 examine laws in thermodynamics and solutions of electrolytes Chapters 6 to 8 look at acid base equilibria and the link between electromagnetic radiation and the structure of atoms Chapters 9 to 11 cover different types of bonding the rates of chemical reactions and the process of adsorption Chapters 12 to 14 present molecular aggregates magnetic resonance spectroscopy and photochemistry and radiation This book is useful to biological scientists for self study and reference With modest additions of mathematical material by the teacher the book should also be suitable for a full year major s course in physical chemistry *Veterinary Anesthesia and Analgesia* Kurt A. Grimm, Leigh A.

Lamont, William J. Tranquilli, Stephen A. Greene, Sheila A. Robertson, 2015-05-11 *Veterinary Anesthesia and Analgesia* the Fifth Edition of Lumb and Jones is a reorganized and updated edition of the gold standard reference for anesthesia and pain management in veterinary patients Provides a thoroughly updated edition of this comprehensive reference on veterinary anesthesia and analgesia combining state of the art scientific knowledge and clinically relevant information Covers immobilization sedation anesthesia and analgesia of companion wild zoo and laboratory animals Takes a body systems approach for easier reference to information about anesthetizing patients with existing conditions Adds 10 completely new chapters with in depth discussions of perioperative heat balance coagulation disorders pacemaker implantation cardiac output measurement cardiopulmonary bypass shelter anesthesia and pain management anesthetic risk assessment principles of anesthetic pharmacology and more Now printed in color with more than 400 images *Atoms, Molecules, and Reactions*

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Physiology and Biophysics: Circulation, respiration, and fluid balance William Henry Howell, Theodore Cedric Ruch, 1973

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Canadian Journal of Chemistry, 1999 General Chemistry Kenneth W. Whitten, Kenneth D. Gailey, Raymond E. Davis, 1988

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