

20.1**DESCRIBING ACIDS AND BASES**
SECTION REVIEW**Objectives**

- Use the properties of acids and bases
- Name an acid or base when given the formula

Key Terms

- acid
- base

Part A: Completion

After this completion exercise, you should know an alternative naming of acids and bases and how they are represented in the periodic table. Which block uses the name "perchloric" which is correct, above gallium, or below bismuth?

An aqueous solution of 1. has acidic properties. Acids contain 2. atoms in 3. the products are 4. and 5. are composed of a metal cation with one or more 6. and a 7.

Acids produced in 8. have when dissolved in water. There is a standard equation for naming acids. For example, HCl is called 9. acid, while HNO₃ is called 10. acid.

Basic properties 11. are produced when they dissolve in water. Bases correspond to these are 12. and named for the anions they are anion for example 13.

14.

Part B: True-False

Correctly decide whether statements are true or false. If true, write T; if false, write F.

15. Acids are protonic acids, containing hydrogen atoms in addition to metal cations.

16. Acids are acids because they contain hydroxide ions.

17. Chlorine forms a covalent hydride when reacting with a base.

20 Describing Acids And Bases Answer Key

**William Henry Howell, Theodore Cedric
Ruch**

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Physical Chemistry and Its Biological Applications Wallace Brey,2012-12-02 Physical Chemistry and Its Biological Applications presents the basic principles of physical chemistry and shows how the methods of physical chemistry are being applied to increase understanding of living systems Chapters 1 and 2 of the book discuss states of matter and solutions of nonelectrolytes Chapters 3 to 5 examine laws in thermodynamics and solutions of electrolytes Chapters 6 to 8 look at acid base equilibria and the link between electromagnetic radiation and the structure of atoms Chapters 9 to 11 cover different types of bonding the rates of chemical reactions and the process of adsorption Chapters 12 to 14 present molecular aggregates magnetic resonance spectroscopy and photochemistry and radiation This book is useful to biological scientists for self study and reference With modest additions of mathematical material by the teacher the book should also be suitable for a full year major s course in physical chemistry *Veterinary Anesthesia and Analgesia* Kurt A. Grimm,Leigh A.

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Physiology and Biophysics: Circulation, respiration, and fluid balance William Henry Howell,Theodore Cedric Ruch,1973

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Canadian Journal of Chemistry ,1999 General Chemistry Kenneth W. Whitten,Kenneth D. Gailey,Raymond E. Davis,1988

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