

Acid-Base Equilibrium Homework Part II

Part A

- Calculate the pH and pOH of a solution containing 7.30 grams of HCl in 1.0 L of solution.

$$7.30 \text{ g} \times \frac{1 \text{ mol}}{36.46 \text{ g}} = 0.200 \text{ mol} / 1.0 \text{ L} = 0.20 \text{ M HCl}$$

Because HCl is a strong acid, it completely dissociates, so $[\text{HCl}] = [\text{H}_3\text{O}^+] = 0.20 \text{ M}$
 $\text{pH} = -\log[0.20] = 0.70$
 $\text{pOH} = 14 - \text{pH} = 13.30$
- Calculate the pH of a 0.10 M nitric acid (HNO_3) solution.

Because HNO_3 is a strong acid, the acid completely dissociates and the concentration of the acid is equal to the concentration of H^+ at equilibrium.
 $[\text{H}_3\text{O}^+] = 0.10 \text{ M}$
 $\text{pH} = -\log(0.10) = 1$
- Calculate the pH of a $5.0 \times 10^{-3} \text{ M}$ solution of sodium hydroxide.

Sodium hydroxide is a strong base, which means it completely dissociates, so $[\text{NaOH}] = [\text{OH}^-]$.
 $[\text{OH}^-] = 5.0 \times 10^{-3}$
 $\text{pOH} = -\log(5.0 \times 10^{-3}) = 1.3$
 $\text{pH} = 14 - 1.3 = 12.7 = 13 \text{ (2 sig figs)}$

Part B

- Decide whether the statements are true about strong acids, weak acids, both, or neither.
 - Always dissociates leaving behind at least some hydronium ion **both**
 - The initial concentration of the acid is equivalent to the final concentration of hydronium ions **strong only**
 - The pH is easily calculated **strong only**
 - The pH must be calculated by the ICE method **weak only**
 - The acid completely dissociates in water **strong only**
 - The equilibrium concentration of undissociated acid is approximately 0 **strong only**
 - K_a is very small **weak only**
- Write the K_a expression for the following acids:
 - $\text{HC}_2\text{H}_3\text{O}_2$

$$\text{HC}_2\text{H}_3\text{O}_2 (\text{aq}) + \text{H}_2\text{O}(\text{l}) \leftrightarrow \text{C}_2\text{H}_3\text{O}_2^- (\text{aq}) + \text{H}_3\text{O}^+ (\text{aq})$$

$$K_a = \frac{[\text{C}_2\text{H}_3\text{O}_2^-][\text{H}_3\text{O}^+]}{[\text{HC}_2\text{H}_3\text{O}_2]}$$
 - H_2CO_3

$$\text{H}_2\text{CO}_3 (\text{aq}) + \text{H}_2\text{O}(\text{l}) \leftrightarrow \text{HCO}_3^- (\text{aq}) + \text{H}_3\text{O}^+ (\text{aq})$$

$$K_a = \frac{[\text{HCO}_3^-][\text{H}_3\text{O}^+]}{[\text{H}_2\text{CO}_3]}$$

Acid Base Homework Answers

James Dowd, MD, Diane Stafford



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Michael B. Smith, 2016-03-09 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts *Organic Chemistry An Acid Base Approach* provides a framework for understanding the subject that goes beyond mere memorization Using several techniques to develop a relational understanding it helps students fully grasp the essential concepts at the root of organic chemistry This new edition was rewritten largely with the feedback of students in mind and is also based on the author's classroom experiences using the first edition Highlights of the Second Edition Include Reorganized chapters that improve the presentation of material Coverage of new topics such as green chemistry Adding photographs to the lectures to illustrate and emphasize important concepts A downloadable solutions manual The second edition of *Organic Chemistry An Acid Base Approach* constitutes a significant improvement upon a unique introductory technique to organic chemistry The reactions and mechanisms it covers are the most fundamental concepts in organic chemistry that are applied to industry biological chemistry biochemistry molecular biology and pharmacy Using an illustrated conceptual approach rather than presenting sets of principles and theories to memorize it gives students a more concrete understanding of the material *The Vitamin D Cure, Revised* James Dowd, MD, Diane Stafford, 2012-05-16 Now updated with the latest research the breakthrough way to lose weight stop pain and prevent disease Completely updated with the latest research *The Vitamin D Cure* tells you all you need to know about this miraculous natural substance today's best way to heal pain prevent disease and improve your mood We now know that adding vitamin D to your daily regimen can net you unbelievable benefits from reducing your chances of having certain kinds of cancer to gaining flexibility and youthful exuberance well into your seventies and beyond And no a dose of daily sunshine doesn't give you enough Groundbreaking research reveals the healing power of this simple readily available supplement Physicians researchers and vitamin D users point to mounting evidence that the simple act of increasing the amount of vitamin D in your body can cure or help treat a remarkable number of diseases and ailments including unwanted pounds high blood pressure back pain muscle cramps obesity cancer and diabetes Leading rheumatologist and researcher Dr James Dowd reveals the causes of vitamin D deficiency and offers a simple five step program that can prevent or alleviate a host of health troubles in a matter of weeks Incorporating the latest diet and lifestyle advice the program helps you harness the power of the sunshine vitamin to transform your life Whether you struggle with arthritis or chronic pain from an old injury or you simply want to lose weight and get back to a body image that reflects your best self *The Vitamin D Cure* can be your key to a better more vibrant you All new recipes and meal plans Up to date information on how vitamin D helps prevent disease and reduce pain New diet and supplement breakthroughs New information from scientific studies a wealth of clinical data and case stories *Chemistry Homework* Frank Schaffer Publications, Joan DiStasio, 1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more *Basic Chemical Principles* Donada Peters, Edward I. Peters, 1988 **Active Learning** Patricia Ann Mabrouk, 2007 This symposium series book focuses on the application of

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