

## ACID & BASE

- Provide optimal cell function
  - Cells secrete **acid** and **bases** to maintain proper pH for enzymes to work.
- They both provide homeostatic regulation of pH of the body's ECF (extracellular fluid)
  - The Carbonic Acid-Bicarb Buffer System = pH homeostasis of blood

### ACID

CO<sub>2</sub> is regulated by the lungs + contributes to acid

- found in the stomach + urine + kidneys \*
  - ex: acid reflux, BPH, kidney failure

### BASE

Bicarb HCO<sub>3</sub> = form of carbon dioxide in our blood

- found in intestines \*
  - Regulated, excreted + reabsorbed in kidneys \*
  - Ex: C-diff → bacteria in colon

## 2 FUNCTIONS

Metabolic + Respiratory

### METABOLIC

primary change is the **concentration of bicarbonate.**

#### METABOLIC ACIDOSIS

bicarb ↓ r/t the amount of acid in the body

- Increased Hydrogen ions
  - CAUSES: kidney failure, DKA, lactic acidosis (shock/sepsis), severe diarrhea
  - put pt. on bicarb drip to increase levels
- Respiratory compensation
  - Increased RR to rid of acid

#### METABOLIC ALKALOSIS

excess bicarb r/t the amount of acid in the body

- Decreased Hydrogen ions
  - CAUSES: loss of gastric fluids, antacid overdose
  - ex: too much NGT suctioning, hypokalemia, vomiting

### RESPIRATORY

primary change is the **concentration of carbonic acid**

#### RESPIRATORY ACIDOSIS

CO<sub>2</sub> is retained + inadequate O<sub>2</sub> exchange → increases amount of carbonic acid in the body

- CAUSES: Hypoventilation
  - ex: Asthma, COPD, Pneumonia
- Decreased RR
  - Sleep apnea, drug overdose, head injuries

Compensation?

- Increasing RR

#### RESPIRATORY ALKALOSIS

Too much CO<sub>2</sub> is lost + carbonic acid levels fall

- Pt has increased RR
  - ex: hyperventilation due to anxiety attack

Compensation?

- slow + shallow breaths

# Acid And Bases Study Guide

**T. W. Graham Solomons**



## Acid And Bases Study Guide:

*Organic Chemistry, 12e Binder Ready Version Study Guide & Student Solutions Manual* T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2016-04-11 This is the Student Study Guide Solutions Manual to accompany Organic Chemistry 12th Edition The 12th edition of Organic Chemistry continues Solomons Fryhle Snyder's tradition of excellence in teaching and preparing students for success in the organic classroom and beyond A central theme of the authors' approach to organic chemistry is to emphasize the relationship between structure and reactivity To accomplish this the content is organized in a way that combines the most useful features of a functional group approach with one largely based on reaction mechanisms The authors' philosophy is to emphasize mechanisms and their common aspects as often as possible and at the same time use the unifying features of functional groups as the basis for most chapters The structural aspects of the authors' approach show students what organic chemistry is Mechanistic aspects of their approach show students how it works And wherever an opportunity arises the authors show students what it does in living systems and the physical world around us *Organic Chemistry, 13e Student Study Guide and Solutions Manual* T. W. Graham Solomons, Craig B. Fryhle, Scott A. Snyder, 2022-05-03 Organic Chemistry Student Study Guide and Solutions Manual 13th Edition offers the full solutions for select exercises from the text **Study Guide for Chemical Principles** Thomas Elliott Taylor, 1979 *Fundamentals of Chemistry, Study Guide* James E. Brady, John R. Holum, 1988-04-20 This Third Edition of the widely used fundamentals textbook for science majors maintains the conversational writing style that made the previous editions so popular while including up to date treatments of important and current topics Emphasizes descriptive chemistry chemical reactions and properties while maintaining a solid treatment of chemical principles Common chemicals are used whenever possible as examples in both theoretical discussions and in problems and exercises Incorporates many pedagogical aids each chapter begins with a brief table of contents and each section begins with a preview of topics covered Chapters include frequent margin comments figures and photographs **Study Guide for Chemical Principles [by] Steven S. Zumdahl** Steven S. Zumdahl, Paul B. Kelter, 1995 The Study Guide reflects the unique problem solving approach taken by the Chemical Principles text The new edition of the Study Guide includes many new worked out examples *Acid-Base Equilibria - Quick Chemistry Review Outline and Handout* E Staff, Acid Base Equilibria Quick Review Outline and Handout for All Students Learn and review on the go Use Quick Review Chemistry Notes to help you learn or brush up on the subject quickly You can use the review notes as a reference to understand the subject better and improve your grades Easy to remember facts to help you perform better Perfect study notes for all high school and college students 10 Pages **Study Guide and Full Solutions Manual** Susan McMurry, John McMurry, 2002-08 Contains a brief overview of every chapter review of skills self tests and the answers and detailed solutions to all end of chapter problems in the textbook *Study Guide [to Accompany] General Chemistry* James E. Brady, 1982 Basic Concepts of Chemistry, Study Guide Leo J. Malone, 2000 The text's three main

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In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Acid And Bases Study Guide**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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