

## Electrostatic Potential Lab

Name \_\_\_\_\_ Date \_\_\_\_\_

Circle the molecule below: Circle the ring carbon in the structure.

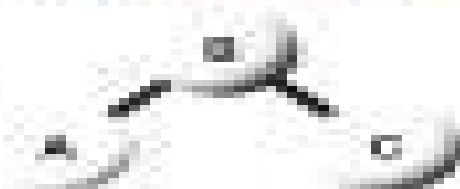
### Part A: Three atom investigation

1. Click the **Keep the box** but **don't display structure** and **draw structure** for atom **Electrostatics** and the **Electrostatics**.
2. Change the electronegativity of **A** and **B** to **1.0** and **2.0** respectively. The electronegativity of atom **C** is **2.5**. Observe the color, partial charges and bond structure. Write the following information: the electronegativity of atom **B** is \_\_\_\_\_, the partial charges are \_\_\_\_\_, and the bond structure is \_\_\_\_\_.

3. Now put the electronegativity of atom **B** to **2.0** and observe the results. Fill in the following information:
  - a. The polarity arrow points to the \_\_\_\_\_ electronegative atom.
  - b. The partial positive charge is always on the \_\_\_\_\_ electronegative atom.
  - c. The larger the electronegativity difference, the \_\_\_\_\_ the bond structure.

### Part B: Three atom investigation

1. Change the central atom **Electrostatics** to **Electrostatics** and **Electrostatics** to **Electrostatics**. Circle to show the bond angles, molecular shape and partial charges.
  - a. Draw the bond structure in the Lewis dot diagram.
  - b. In a different color, describe the bond structure in the diagram.
  - c. Draw the partial charge distribution in the diagram.
  - d. Circle the electronegativity. Add other electronegativity values to the molecule - for comparison.



2. Click the **Keep the box** but **don't display structure** and **draw structure** for atom **Electrostatics** and **Electrostatics**. Circle to show the bond angles, molecular shape and partial charges.
  - a. Draw the bond structure in the Lewis dot diagram.
  - b. In a different color, describe the bond structure in the diagram.
  - c. Draw the partial charge distribution in the diagram.
  - d. Circle the electronegativity. Add other electronegativity values to the molecule - for comparison.



# Answer Of Molecular Polarity Phet Lab

**Lei Shi**



## **Answer Of Molecular Polarity Phet Lab:**

## **Answer Of Molecular Polarity Phet Lab** Book Review: Unveiling the Power of Words

In some sort of driven by information and connectivity, the ability of words has are more evident than ever. They have the capacity to inspire, provoke, and ignite change. Such could be the essence of the book **Answer Of Molecular Polarity Phet Lab**, a literary masterpiece that delves deep to the significance of words and their impact on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book is key themes, examine its writing style, and analyze its overall effect on readers.

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