

- ii. Ternary acids - just like binary acids, ternary acids must always contain hydrogen. These acids also contain a polyatomic ion.

Polyatomic ion	Name of ion	Polyatomic ion	Name of ion
$\text{OH}^-$	hydroxide	$\text{PO}_4^{3-}$	phosphate
$\text{O}_2\text{H}^-$	peroxide	$\text{CO}_3^{2-}$	carbonate
$\text{C}_2\text{H}_5\text{O}_2^-$	ethoxide	$\text{CH}_3\text{CO}_2^-$	acetate
$\text{SO}_3^{2-}$	sulfite	$\text{CH}_3\text{COO}^-$	acetate
$\text{SO}_4^{2-}$	sulfate	$\text{ClO}_4^-$	chlorate

Ternary acids are three element acids. Most ternary acids are **oxyacids** containing hydrogen, oxygen and other atoms.

#### Activity

- Name the seven common binary acids.
- Change the names according to one of the following:

Acid ending	for acid, replace with
-ic	-ous acid
-ous	-ic acid

Formula	composition	Ternary compound name	Ternary acid name
$\text{H}_2\text{SO}_4$	$\text{SO}_4^{2-}$	hydrogen sulfate	Sulfuric acid
$\text{H}_2\text{SO}_3$	$\text{SO}_3^{2-}$	hydrogen sulfite	Sulfurous acid
$\text{H}_2\text{CO}_3$	$\text{CO}_3^{2-}$	hydrogen carbonate	Carbonic acid
$\text{H}_2\text{CO}_4$	$\text{CO}_4^{2-}$	hydrogen bicarbonate	Bicarbonate acid
$\text{H}_2\text{NO}_2$	$\text{NO}_2^-$	nitrogen dioxide	Nitric acid
$\text{H}_2\text{NO}_3$	$\text{NO}_3^-$	nitrogen trioxide	Nitrous acid
$\text{H}_2\text{NO}_4$	$\text{NO}_4^{2-}$	nitrogen tetroxide	Nitrogenic acid

## Answers For Writing Ternary Formulas

**A Loxley**

## Answers For Writing Ternary Formulas:

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the constants a h and k affect the graph of the quadratic function  $g(x) = a(x - h)^2 + k$

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web when a quadratic function is represented in vertex form following points are to be noted if  $h > 0$  graph shifts right by  $h$  units if  $h < 0$  graph shifts left by  $|h|$  units if  $k > 0$  graph shifts upwards by  $k$  units if  $k < 0$  graph shifts downwards by  $|k|$  units h

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