

Name: Ahmed Faizan Date: 12 May Period:

Specific Heat and Heat Capacity Worksheet

DIRECTIONS: Use $q = (m)(C_p)(\Delta T)$ to solve the following problems. Show all work and units.

Ex: How many joules of heat are needed to raise the temperature of 10.0 g of aluminum from 22°C to 55°C, if the specific heat of aluminum is 0.90 J/g°C?

$$\begin{aligned} q &= ? & \Delta t &= 55 - 22 = 33^\circ\text{C} & q &= m s \Delta t \\ m &= 10\text{g} & s &= 0.9 \text{ J/g}^\circ\text{C} & &= 10 \times 0.9 \times 33 \\ & & & & &= \underline{297 \text{ J}} \end{aligned}$$

1. The temperature of 335 g of water changed from 24.5°C to 26.4°C. How much heat did this sample absorb? C for water = 4.18 J/g°C

$$\begin{aligned} \Delta t &= 26.4 - 24.5 = 2 & q &= m s \Delta t \\ s &= 4.18 \text{ J/g}^\circ\text{C} & &= 335 \times 4.18 \times 2 \\ m &= 335\text{g} & &= \underline{2800.6 \text{ J}} \end{aligned}$$

2. How much heat in kilojoules has to be removed from 225g of water to lower its temperature from 25.0°C to 10.0°C?

$$\begin{aligned} \Delta t &= 25 - 10 = 15 & q &= m s \Delta t \\ m &= 225\text{g} & &= 225 \times 4.18 \times 15 \\ s &= 4.18 \text{ J/g}^\circ\text{C} & &= \underline{14107.5 \text{ J}} \\ 1\text{J} &= 1 \times 10^{-3} \text{ KJ} & \therefore &= \underline{14.1075 \text{ KJ}} \end{aligned}$$

3. To bring 1.0kg of water from 25°C to 99°C takes how much heat input?

$$\begin{aligned} m &= 1\text{kg} = 1000\text{g} & q &= 1000 \times 4.18 \times 74 \\ \Delta t &= 99 - 25 = 74 & q &= \underline{309320 \text{ J}} \\ s &= 4.18 \text{ J/g}^\circ\text{C} & & \end{aligned}$$

An insulated cup contains 75.0g of water at 24.00°C. A 26.00g sample of metal at 82.25°C is added. The final temperature of the water and metal is 28.34°C. What is the specific heat of the metal?

Imp

$$\begin{aligned} (q_{\text{water}} &= -q_{\text{metal}}) & (C = m s) &\rightarrow s = \frac{C}{m} \\ q_{\text{water}} &= 75 \times 4.18 \times (28.34 - 24) & q &= m s \Delta t \\ &= 1252 & 1252 &= 26 \times s \times (82.25 - 28.34) \\ & & & 1252 = 26 \times 53.91 \times s \\ & & & s = \underline{0.893 \text{ J/g}^\circ\text{C}} \end{aligned}$$

5. A calorimeter has a heat capacity of 1265 J/°C. A reaction causes the temperature of the calorimeter to change from 22.34°C to 25.12°C. How many joules of heat were released in this process?

$$\begin{aligned} C &= 1265 \text{ J/}^\circ\text{C} \\ \Delta t &= 25.12 - 22.34 = 2.78 \\ q &= ? \\ q &= C \Delta t \\ &= 1265 \times 2.78 \\ &= \underline{3516.7 \text{ J}} \end{aligned}$$

Answers For Specific Heat Cpo Science

Julia Schneider



Answers For Specific Heat Cpo Science:

Nuclear Science Abstracts ,1975-05 *Boundary and Mixed Lubrication: Science and Applications* D. Dowson,2002-07-10 Session headings Keynote address and invited paper Boundary lubrication 1 Boundary lubrication 2 Mixed lubrication Lubricants 1 Lubricants 2 Nano boundary lubrication and molecular dynamics Coatings Dry wear Lubricated wear Bio tribology total replacement hip joints Elastohydrodynamic lubrication Friction 1 Bearings Friction 2 Mechanical systems **Canadian Journal of Chemistry** ,1975 **Numerical Methods in Laminar and Turbulent Flow** Cedric Taylor,B. A. Schrefler,1981 *Theoretical Chemical Engineering Abstracts* ,1970 **Interdisciplinary Finite Element Analysis** John Fredrick Abel,Tadahiko Kawai,Shan-Fu Shen,1981 Journal of the Aero/space Sciences ,1960 *Monthly Weather Review* ,1985 *Science Citation Index* ,1994 Vols for 1964 have guides and journal lists **Physics Briefs** ,1992 Dissertation Abstracts International ,1985-08 Science Abstracts ,1949 **Who's Who in Science and Engineering 2008-2009** Who's Who Marquis,Marquis Who's Who,2007-12 **EPA Publications Bibliography, 1984-1990: Report summaries** United States. Environmental Protection Agency,1990 **EPA Publications Bibliography, 1984-1990: Report summaries** ,1990 **Literary Concordances** Trevor Howard Howard-Hill,1979 All problems likely to be encountered by anyone who intends to prepare a literary concordance are discussed on a practical level although there is substantial examination of more advanced concordancing techniques which the computer makes it possible to adopt Although the emphasis is on works in English the structural principles which are analysed in the book can be applied readily to works in other languages *EPA Publications Bibliography* ,1983 Consolidates into one publication all reports previously published in the EPA publications bibliography quarterly abstract bulletin from its inception in 1977 Foreword 1977 1983 **Specific Heat** Yeram Sarkis Touloukian,1976 Specific Heat Capacity ,2006 **Specific Heat of Solutions which are Not Electrolytes** William Francis Magie,1899

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web recognize the anatomical structures explain physiological functions and recognize and explain the principle of homeostasis applied to the integumentary nervous endocrine muscular and skeletal systems

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