

# Conjugate Acid-Base Pairs

Fill in the missing information in the two tables about conjugate acids and bases.

Acid	Base	Conjugate Acid	Conjugate Base	Equation
$\text{HNO}_2$	$\text{H}_2\text{O}$			$\text{HNO}_2 + \text{H}_2\text{O} \rightarrow \text{NO}_2^- + \text{H}_3\text{O}^+$
$\text{H}_2\text{O}$	$\text{F}^-$	$\text{HF}$	$\text{OH}^-$	
				$\text{NH}_3 + \text{HCN} \rightarrow \text{NH}_4^+ + \text{CN}^-$
		$\text{H}_2\text{O}$	$\text{ClO}_3^-$	
$\text{HSO}_4^-$	$\text{PO}_4^{3-}$			
				$\text{H}_2\text{O} + \text{S}^{2-} \rightarrow \text{HS}^- + \text{OH}^-$
$\text{HCO}_2\text{H}$	$\text{OH}^-$			

Acid	Conjugate Base	Equation
$\text{H}_3\text{PO}_4$		
	$\text{F}^-$	
	$\text{NO}_3^-$	
$\text{H}_2\text{PO}_4^-$		
$\text{H}_2\text{O}$		
	$\text{SO}_4^{2-}$	
$\text{HPO}_4^{2-}$		

# Answers For Conjugate Acid Base Pairs

**Ronald James Gillespie**



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