

13. (4 pts) What is an ionic compound?
a. A compound that will dissolve in water
b. Two nonmetal atoms bonded together
c. Two metal atoms bonded together
d. A metal and a nonmetal atom bonded together
e. None of the above

14. (4 pts) What type of compound uses the naming system consisting of prefixes like *mono-*, *di-*, and *tri-*?
a. Ionic
b. Covalent
c. Both a and b
d. Neither a nor b

15. (4 pts) What type of compound uses a naming system that has *-ide* as the ending on the nonmetals?
a. Ionic
b. Covalent
c. Both a and b
d. Neither a nor b

16. (4 pts) How many total atoms are in this molecule: $\text{C}_6\text{H}_{12}\text{OH}$?
a. 3
b. 4
c. 19
d. 20
e. None of the above

17. (4 pts) You are told that a molecule contains a total of 17 atoms. If the formula is $\text{C}_x\text{H}_{10}\text{Cl}_3$, what must x be?
a. 1
b. 2
c. 3
d. 4
e. None of the above

18. (4 pts) Which of the following compounds would conduct electricity when dissolved in water? $\text{C}_6\text{H}_5\text{ClF}$ $\text{C}_6\text{H}_5\text{CH}_3$ CS_2 BaSO_4
a. $\text{C}_6\text{H}_5\text{ClF}$
b. $\text{C}_6\text{H}_5\text{CH}_3$
c. CS_2
d. BaSO_4
e. CS_2 and BaSO_4

Answers For Chemistry Module 13

Camilla Mei-Fang Kao

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Nelson Modular Science Paul Collinson,Paul Collison,2005-03 The Nelson Modular Science series is made up of three books divided into single double and triple award modules presented in an accessible format Book 1 covers the six single award and one coursework modules Book 2 contains six double award modules and Book 3 covers the six triple award modules Each module is covered in self contained units This teacher s file includes practical support sheets and addresses Sc1 investigations Works sheets are provided to integrate the use of ICT throughout science Additional GCSE style questions and modular tests should enhance learning and recall of information

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Roger Norris,2015-10-08 The Cambridge IGCSE Chemistry Revision Guide supports students through their course containing specifically designed features to help students apply their knowledge as they prepare for assessment This Revision Guide offers support for students as they prepare for their Cambridge IGCSE Chemistry 0620 exams Containing up to date material that matches the syllabus for examination from 2016 and packed full of guidance such as Worked Examples Tips and Progress Check questions throughout to help students to hone their revision and exam technique and avoid common mistakes These features have been specifically designed to help students apply their knowledge in exams Written in a clear and straightforward tone this Revision Guide is perfect for international learners

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Study,1980 Reviewing Chemistry Peter E. Demmin,1989 *Educational Technology* ,1966 **Biology/science Materials** Carolina Biological Supply Company,1991 *Structure, Function and Engineering of Modular Polyketide Synthases* Camilla Mei-Fang Kao,1997

Answers For Chemistry Module 13 Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has been more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Answers For Chemistry Module 13**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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