

Nuclear Chemistry (used in activities)

Model: Models represent the three isotopes of carbon:



Critical Thinking Questions

1. How many protons are in carbon-12? $\frac{6}{12}$ How many neutrons are in carbon-12? $\frac{6}{12}$
2. How many electrons are in carbon-12? $\frac{6}{12}$ How many neutrons are in carbon-12? $\frac{6}{12}$
3. How many protons are in carbon-13? $\frac{6}{13}$ How many neutrons are in carbon-13? $\frac{7}{13}$
4. How is carbon-14 different from carbon-12? What is different among isotopes?
(Note: in previous 3 questions)
5. What does the subscript represent in all three models (carbon-12, carbon-13, carbon-14)?

Information: Nuclear reactions and binding/decay:

- A nuclear reaction is a change in the composition of the nucleus of an atom. That is not necessarily accompanied by chemical reaction, and does not depend on what substances are there except for the.
- There are three types of nuclear reactions: fusion, fission, and radioactive decay. Fusion is the joining of two light nuclei, such as the stars where the sun and fusion. Fission is the splitting of a heavy nucleus, such as the bomb where the atom is split. Radioactive decay is the process of a nucleus changing into a more stable nucleus.
- Some reactions are exothermic, meaning that they release energy. Some reactions are endothermic, meaning that they absorb energy. Some reactions are both, meaning that they release energy and absorb energy.

Table 1: Types of nuclear reactions and their products

Type of Reaction	Symbol	Mass Number	Charge	Energy Released/Absorbed	Products	Isotopes Produced
Fusion	α , β	4	2+	very high	Helium-4	Helium-4 (stable)
Fission	α , β	4	2+	very high	Helium-4	Helium-4 (stable)
Radioactive Decay	α , β	4	2+	very high	Helium-4	Helium-4 (stable)
Exothermic	α , β	4	2+	very high	Helium-4	Helium-4 (stable)
Endothermic	α , β	4	2+	very high	Helium-4	Helium-4 (stable)

Answers Chemactivity 6

Lauren Gardner



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Introductory Chemistry Michael P. Garoutte, Ashley B. Mahoney, 2015-08-10 The ChemActivities found in Introductory Chemistry A Guided Inquiry use the classroom guided inquiry approach and provide an excellent accompaniment to any one semester Introductory text Designed to support Process Oriented Guided Inquiry Learning POGIL these materials provide a variety of ways to promote a student focused active classroom that range from cooperative learning to active student participation in a more traditional setting General, Organic, and Biological Chemistry Michael P. Garoutte, 2014-02-24 Classroom activities to support a General Organic and Biological Chemistry text Students can follow a guided inquiry approach as they learn chemistry in the classroom General Organic and Biological Chemistry A Guided Inquiry serves as an accompaniment to a GOB Chemistry text It can suit the one or two semester course This supplemental text supports Process Oriented Guided Inquiry Learning POGIL which is a student focused group learning philosophy of instruction The materials offer ways to promote a student centered science classroom with activities The goal is for students to gain a greater understanding of chemistry through exploration *Organic Chemistry, a Guided Inquiry* Andrei Straumanis, 2004 Merrill Chemistry Robert C. Smoot, Smoot, Richard G. Smith, Jack Price, 1998 **Prentice Hall Chemistry**, 2000 **Chemistry** Richard S. Moog, John J. Farrell, 2008-01-09 The ChemActivities found in Chemistry A Guided Inquiry Fourth Edition use the classroom guided inquiry approach and provide an excellent accompaniment to Spencer s Chemistry Structure and Dynamics Fourth Edition or any other General Chemistry text Designed to support Process Oriented Guided Inquiry Learning POGIL these materials provide a variety of ways to promote a student focused active classroom that range from cooperative learning to active student participation in a more traditional setting They are designed to train students to use and analyze data figures and text to deduce chemical concepts **Process Oriented Guided Inquiry Learning (POGIL)** Richard Samuel Moog, 2008 POGIL is a student centered group learning pedagogy based on current learning theory This volume describes POGIL s theoretical basis its implementations in diverse environments and evaluation of student outcomes **Chemists' Guide to Effective Teaching** Norbert J. Pienta, Melanie M. Cooper, Thomas J. Greenbowe, 2005 For courses in Methods of Teaching Chemistry Useful for new professors chemical educators or students learning to teach chemistry Intended for anyone who teaches chemistry or is learning to teach it this book examines applications of learning theories presenting actual techniques and practices that respected professors have used to implement and achieve their goals Each chapter is written by a chemist who has expertise in the area and who has experience in applying those ideas in their classrooms This book is a part of the Prentice Hall Series in Educational Innovation for Chemistry

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