

KEV

Electron Review Questions

Using the equations, solve the following:

$$c = \lambda \nu \quad c = \text{speed of light } 3.0 \times 10^8 \text{ m/sec} \quad \nu = \text{frequency} \quad \lambda = \text{wavelength}$$

$$E = h\nu \quad E = \text{energy in joules} \quad h = \text{planck constant } 6.626 \times 10^{-34} \quad \nu = \text{frequency}$$

1. What is the frequency of a wave that has a wavelength of 4×10^7 meters?
 $\nu = \frac{c}{\lambda} = \frac{3.0 \times 10^8 \text{ m/sec}}{4 \times 10^7 \text{ m}} = 7.50 \times 10^{14} \text{ s}^{-1}$

2. What is the energy of this wave?

$$E = h\nu = 6.626 \times 10^{-34} \times 7.5 \times 10^{14} = 4.97 \times 10^{-19} \text{ J}$$

3. List 5 forms of electromagnetic radiation and what is the lowest energy color of light? Which color of light has the highest energy?

Radio waves, microwaves, visible light, UV, x-rays, gamma

4. What is the relationship between frequency and wavelength?

IT is inversely proportional. i.e., as wavelength decreases frequency increases it

5. What is the relationship between energy and frequency?

IT is directly proportional. As frequency increases, energy increases

6. Where are the electrons located in the atom?

well around the nucleus in the electron cloud

7. Define:

a) Ground State - lowest energy state of an atom, a neutral atom

b) Excited State - state when atom gains energy

8. What is a bright-line or line-emission spectrum of an atom?

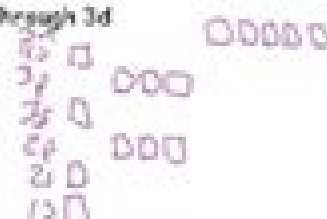
a series of specific frequencies and wavelengths of visible light that appear when an element's light is passed through a prism

9. The principle that says no two electrons can have the same set of quantum numbers is Pauli Exclusion Principle

10. The principle that says electrons will orbit the lowest energy orbitals first is called Aufbau principle

11. The rule that says electrons will occupy orbitals of equal energy alone, before pairing up is called Hund's rule

12. Draw the Aufbau Diagram through 3d



13. Name where the blocks located on the periodic table:

a) s block? 1st & 2nd columns

b) p block? right 6 columns

c) d block? in the middle between s and p (8 gray boxes)

d) f block? two rows at the bottom

14. What is the shape of the p orbital? s orbital?

dumbbell, spherical

15. Write the electron configuration using orbital notation for sulfur:



16. Write the electron configuration for Cadmium using electron configuration notation.



17. Write the electron configuration for lead using noble gas notation:



18. How many valence electrons does sulfur have?



19. Why are copper and chromium exceptions to the Aufbau principle?

they borrow 1e⁻ from 4s and halfing fill 3d for Cr and completely fill 3d for Cu

20. Write the electron configuration for the element Sb using the Noble gas abbreviation.



Answer Key Unit 3 Review Electrons

Arshad Iqbal



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The Top Books of the Year Answer Key Unit 3 Review Electrons The year 2023 has witnessed a noteworthy surge in literary brilliance, with numerous compelling novels enthralling the hearts of readers worldwide. Lets delve into the realm of popular books, exploring the captivating narratives that have enthralled audiences this year. Answer Key Unit 3 Review Electrons : Colleen Hoover's "It Ends with Us" This poignant tale of love, loss, and resilience has captivated readers with its raw and emotional exploration of domestic abuse. Hoover expertly weaves a story of hope and healing, reminding us that even in the darkest of times, the human spirit can prevail. Uncover the Best : Taylor Jenkins Reids "The Seven Husbands of Evelyn Hugo" This spellbinding historical fiction novel unravels the life of Evelyn Hugo, a Hollywood icon who defies expectations and societal norms to pursue her dreams. Reids absorbing storytelling and compelling characters transport readers to a bygone era, immersing them in a world of glamour, ambition, and self-discovery. Discover the Magic : Delia Owens "Where the Crawdads Sing" This captivating coming-of-age story follows Kya Clark, a young woman who grows up alone in the marshes of North Carolina. Owens spins a tale of resilience, survival, and the transformative power of nature, captivating readers with its evocative prose and mesmerizing setting. These bestselling novels represent just a fraction of the literary treasures that have emerged in 2023. Whether you seek tales of romance, adventure, or personal growth, the world of literature offers an abundance of engaging stories waiting to be discovered. The novel begins with Richard Papen, a bright but troubled young man, arriving at Hampden College. Richard is immediately drawn to the group of students who call themselves the Classics Club. The club is led by Henry Winter, a brilliant and charismatic young man. Henry is obsessed with Greek mythology and philosophy, and he quickly draws Richard into his world. The other members of the Classics Club are equally as fascinating. Bunny Corcoran is a wealthy and spoiled young man who is always looking for a good time. Charles Tavis is a quiet and reserved young man who is deeply in love with Henry. Camilla Macaulay is a beautiful and intelligent young woman who is drawn to the power and danger of the Classics Club. The students are all deeply in love with Morrow, and they are willing to do anything to please him. Morrow is a complex and mysterious figure, and he seems to be manipulating the students for his own purposes. As the students become more involved with Morrow, they begin to commit increasingly dangerous acts. The Secret History is a brilliant and thrilling novel that will keep you wondering until the very end. The novel is a warning tale about the dangers of obsession and the power of evil.

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