

Analytical Chemistry 3rd Long Exam

PART 1. MULTIPLE-CHOICE: Choose the letter that corresponds to the **BEST** answer. Write **E** when the answer is **NOT** in the choices. (2 pts)

1. Transfer of players from one country to another

2. What is the mole ratio of Ce^{IV} to NO_3^- for the oxidation of nitrite to nitrate using Ce^{IV} in the presence of an acid?



a. 1:2
b. 2:1
c. 1:1
d. 2:2

3. How many mole(s) of electron (e⁻) are produced in the oxidation of sodium stannite (Na_2SnO_3) to SnO_4^{2-} and Na^+ under acidic conditions?

4. Ascorbic acid ($\text{NaC}_6\text{H}_4\text{O}_6$) was added in the determination of the dissolved oxygen in water samples to remove interferences caused by the presence of what kind?

- a. chloride
- b. fluoride
- c. nitrite
- d. nitrate

5. Which of the following measures should not

- I. Prepared KMnO_4 solutions should be stored in amber bottles
- II. KMnO_4 solutions are boiled to remove CO_2 which makes the solution basic
- III. MnO_2 is the product of the degradation of KMnO_4 .
- IV. KMnO_4 solutions are filtered to remove manganese(IV) oxide

6. What is the oxidation state of carbon in acrylic acid?

a. +2	c. +6
b. +4	d. +7

7. The complex formed between Cu(II) and 1,10-phenanthroline has a molar absorptivity of 7000 $L \cdot cm^{-1} \cdot mol^{-1}$ at 435 nm, the wavelength of maximum absorption. What is the absorbance of a 6.17×10^{-4} M solution of the complex when measured in a 1.00-cm cell at 435 nm?

a. 0.0401 b. 0.0402 c. 0.401 d. 0.402

B. In the presence of iodine, thiourea ion is oxidized quantitatively to tetraiodoanion. What is the chemical formula of tetraiodoanion and

a. 5.0%
b. 5.0%

Q. In winter situation, the fog is produced from

c. oxidation of I^- to I_2
 d. reduction of I_2 to I^-
 e. oxidation of MnO_4^- to MnO_4^{2-}
 f. reduction of MnO_4^- to MnO_4^{2-}

10. A component of a UV-Vis spectrophotometer that restricts the radiation being measured to a narrow band that is absorbed or emitted by the sample.

a. light source	c. radioactive tracer
b. detector	d. microbeam selection

11. Component of a UV-Vis instrument that operates by blocking or absorbing oil but a saturated bond of radiation.

- a. light source
- b. detector
- c. radiation filter
- d. wavelength selector

12. Oxidation of VO_2^+ using permanganate or cerium as standard solution produces $\text{VO}_2\text{O}_2\text{Cl}_2$, what is the oxidation state of vanadium in $\text{VO}_2\text{O}_2\text{Cl}_2$?

Analytical Chemistry Exam Questions

**Douglas A. Skoog, Donald M. West, F.
James Holler**

Analytical Chemistry Exam Questions:

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