

Analytical Chemistry 3rd Long Exam

PART 1. MULTIPLE CHOICE: Choose the letter that corresponds to the **BEST** answer. Write **E** when the answer is **NOT** in the choices. (23 pts)

1. Transfer of electrons from one molecule to another

- a. precipitation rate
- b. soil C
- c. NPP
- d. soil respiration rate

2. What is the mole ratio of Ce^{IV} to NO_2^- for the oxidation of nitrite to nitrate using Ce^{III} in the presence of an acid?



a. 1:2
b. 2:1
c. 1:1
d. 2:2

3. How many mole(s) of electron (are) produced in the oxidation of sodium stannite (Na_2SnO_3) to SnO_4^{2-} and Na^+ under acidic conditions?

4. Ascorbic acid ($\text{NaC}_6\text{H}_4\text{O}_6$) was added in the determination of the dissolved oxygen in water samples to remove interferences caused by the presence of chlorinated

- a. chloride
- b. fluoride
- c. nitrite
- d. nitrate

5. Which of the following measures should not

- I. Prepared KMnO_4 solutions should be stored in amber bottles
- II. KMnO_4 solutions are boiled to remove CO_2 which makes the solution basic
- III. $\text{Mn}(\text{II})$ is the product of the degradation of KMnO_4 .
- IV. KMnO_4 solutions are filtered to remove manganese(IV) oxide

8. What is the oxidation state of carbon in acrylic acid?

$$a_1 + 2 \quad a_2 + 6$$

7. The complex formed between Cu(II) and 1,10-phenanthroline has a molar absorptivity of 7000 $\text{L cm}^{-1} \text{ mol}^{-1}$ at 435 nm, the wavelength of maximum absorption. What is the absorbance of a 6.17×10^{-4} M solution of the complex when measured in a 1.00-cm cell at 435 nm?

a. 0.0411
b. 0.0412
c. 0.4311
d. 0.4312

8. In the presence of iodine, thiourea ion is oxidized quantitatively to tetraiodo-urea ion. What is the chemical formula of tetraiodo-urea ion?

a. 5.0%
b. 5.0%

Q. In winter season, the fog is produced from

c. oxidation of I₂ to I₃
d. reduction of MnO₄ to Mn²⁺
e. reduction of I₂ to I₃
f. reduction of MnO₄ to Mn²⁺ no electrons

10. A component of a UV-Vis spectrophotometer that receives the radiation being measured is a narrow band that is absorbed or emitted by the sample.

a. light source	c. radioactive filter
b. detector	d. wavelength selector

11. Component of a UV-Vis instrument that operates by blocking or absorbing all but a specified band of radiation.

a. light source	c. radiation filter
b. detector	d. wavelength selector

12. Oxidation of VO^{2+} using permanganate or cerium as standard solution produces $\text{VO}(\text{OAc})_4$. What is the oxidation state of vanadium in $\text{VO}(\text{OAc})_4$?

Analytical Chemistry Exam 1 Multiple Choice

Mary Millhollen, Richard H. Langley

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