

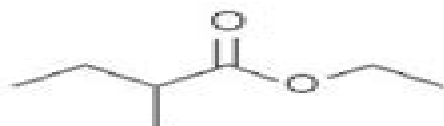
15. Predict which of the following compounds will have a lower boiling point than 1-butanol.

- a. 1-butanol
- b. 2-butanol
- c. 1-butene
- d. butanoic acid

16. Predict the product of the following reaction.  $\text{PhCH}_2\text{OH} + \text{PCC}$  (pyridinium chlorochromate) in methylene chloride  $\rightarrow$

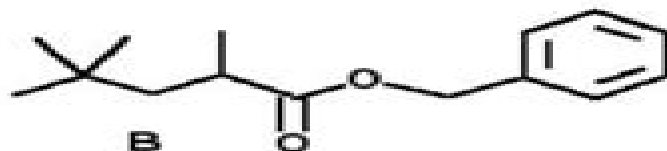
- a. benzophenone
- b. benzoic acid
- c. benzaldehyde
- d. benzyl chloride

17. What is a proper name for the following ester?



- a. 2-butyl ethylester
- b. ethyl 2-pentanoate
- c. ethyl 2-methylbutanoate
- d. methylbutanoic ethyl ester

18. The correct IUPAC name for ester B is



- a. phenylmethyl-2,4,4-trimethylpentanoate
- b. benzyl 2-methyl-4,4-dimethylpentanoate
- c. benzyl 2,4,4-trimethylpentanoate
- d. benzyl 2,4,4-trimethylbutanoate

# Advanced Organic Chemistry Review Questions

**S Nieto**



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student's understanding of basic concepts and to illustrate how structural changes influence mechanism and reactivity

Organic Chemistry in Medicine Budin Michov, 2025-01-24 This book is intended for students in medicine pharmacy and dentistry physicians dentists pharmacists biochemists and more It describes organic compounds that are of importance for medicine heterocycles alkaloids carbohydrates lipids proteins enzymes nucleic acids and more Organic chemistry plays a pivotal role in medicine and in developing synthesizing and understanding pharmaceutical compounds It focuses on carbon containing compounds which form the backbone of organic molecules Functional groups within organic molecules modulate the biochemical properties of organic compounds like stability solubility and activity Organic chemistry is used for designing drug formulations based on disease mechanisms Short biographies of chemists and scientists which have rendered services to general and inorganic chemistry in medicine are given in a person index The Educational Times, and Journal of the

College of Preceptors, 1898 A Foundation Course for College Organic Chemistry B. S. Balaji, 2024-08-22 To understand and improve the underlying principles that govern how organic reactions occur A Foundation Course for College Organic Chemistry follows a brick by brick building approach Emphasis is given to interrelating experimental facts and findings with predictions mechanism and inferences results Discussions focus on clarifying how complex organic reactions occur which is based on electronegativity differences movement of electrons through framework or bonds and addition or removal of atoms hydrogen halogens or groups hydroxy amino The book begins with simple rules governing the deconstruction of reactions and applies them to explain how esterification amide and cyanide hydrolysis reactions proceed The importance of stereochemistry used in drug development biology and medicine aromatic electrophilic and nucleophilic substitutions reaction kinetics and dynamics is explained with suitable examples Features A systematic and structured approach is used to study all aspects of reactive intermediates generation structure geometry and reactions of carbocations carbanions and carbon free radicals This book incorporates scientific methods to deduce reaction mechanisms with simple and relevant explanations and limitations A proper explanation is given to understand the influence of functional groups on the stability and reactivity of intermediates pKa HSAB principles structure activity relations and how these can be exploited in organic chemistry Information is presented in an accessible way for students teachers researchers and scientists **Educational**

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