

## Experiment 3

### Advance Study Assignment: Resolution of Matter into Pure Substances, II. Fractional Crystallization

1. Using Figure 3.1, determine

- a. the number of grams of  $\text{KNO}_3$  that will dissolve in 100 g of  $\text{H}_2\text{O}$  at  $100^\circ\text{C}$ .

240 g  $\text{KNO}_3$

- b. the number of grams of water required to dissolve 25 g of  $\text{KNO}_3$  at  $100^\circ\text{C}$ . (Hint: your answer to Part a gives you the needed conversion factor for g  $\text{KNO}_3$  to g  $\text{H}_2\text{O}$ .)

10.41 g  $\text{H}_2\text{O}$

- c. the number of grams of water required to dissolve 2.0 g  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  at  $100^\circ\text{C}$ .

1.11 g  $\text{H}_2\text{O}$

- d. the number of grams of water required to dissolve a mixture containing 25 g  $\text{KNO}_3$  and 2.0 g  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$ , assuming that the solubility of one substance is not affected by the presence of another.

11.52 g  $\text{H}_2\text{O}$

2. To the solution in Problem 1d at  $100^\circ\text{C}$ , 10 g of water are added, and the solution is cooled to  $0^\circ\text{C}$ .

- a. How much  $\text{KNO}_3$  remains in solution?

1 g  $\text{KNO}_3$

- b. How much  $\text{KNO}_3$  crystallizes out?

24 g  $\text{KNO}_3$

- c. How much  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$  crystallizes out?

2 g  $\text{CuSO}_4 \cdot 5\text{H}_2\text{O}$

- d. What per cent of the  $\text{KNO}_3$  in the sample is recovered?

4 %

# Advance Study Assignment Answers Experiment 1 Chemistry

**SB Merriam**



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