

Chapter 1: Chemical Foundations

1. Which of the following is an example of a quantitative observation?

A) The piece of metal is longer than the piece of wood.
B) Solution 1 is much darker than solution 2.
C) The liquid in beaker A is blue.
D) The temperature of the liquid is 60°C.
E) At least two of the above (A-D) are quantitative observations.

ANS: D

DIF: Easy

REF: 1.2

KEY: Chemistry | general chemistry | general concepts | scientific method

MSC: Conceptual

2. A quantitative observation

A) contains a number and a unit.
B) does not contain a number.
C) always makes a comparison.
D) must be obtained through experimentation.
E) is none of these.

ANS: A

DIF: Easy

REF: 1.2

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3. Generally, observed behavior that can be formulated into a statement, sometimes mathematical in nature, is called a(n)

A) observation.
B) measurement.
C) theory.
D) natural law.
E) experiment.

ANS: D

DIF: Easy

REF: 1.2

KEY: Chemistry | general chemistry | general concepts | scientific method

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4. The statement "The total mass of materials is not affected by a chemical change in those materials" is called a(n)

A) observation.
B) measurement.
C) theory.
D) natural law.
E) experiment.

ANS: D

DIF: Easy

REF: 1.2

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5. A chemical theory that has been known for a long time becomes a law.

Answers To Zumdahl 4th Edition Ch 7

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