

Unit Test 1: Matter, Chemical Trends, & Chemical Bonding**Multiple Choice (10 marks) (KU: 3, C: 3, I: 2)**

Identify the choice that best completes the statement or answers the question. Place your answers in the key below.

1. _____ 2. _____ 3. _____ 4. _____ 5. _____
 6. _____ 7. _____ 8. _____ 9. _____ 10. _____

- The element found in the periodic table in Group 6 and Period 5 is
 - tungsten, W
 - molybdenum, Mo
 - tantalum, Ta
 - tellurium, Te
- Which of the following groups is known for having both metallic and non-metallic properties?
 - alkali metals
 - alkali earth metals
 - transition metals
 - metalloids
- How many electrons does a strontium atom have in its outer shell?
 - 1
 - 2
 - 7
 - 8
- Which particles are responsible for the similar chemical and physical properties of elements in the same group?
 - protons and neutrons
 - number of energy shells
 - valence electrons
 - isotopes
- Which of the following chemical formulas most likely represents a pure substance containing ionic bonds?
 - Cl_2
 - SO_2
 - CO_2
 - BaO
- Consider the equation $\text{X}_{(g)} + e^- \rightarrow \text{X}^- + \text{energy}$. The "energy" term in the equation represents
 - electronegativity
 - electron affinity
 - ionization energy
 - nuclear charge
- Elements A, B, C, and D (found in Groups 1–17) have atomic radii of $265 \mu\text{m}$, $160 \mu\text{m}$, $185 \mu\text{m}$, and $175 \mu\text{m}$, respectively. Which element will most likely have the lowest ionization energy?
 - A
 - B
 - C
 - D
- List the following elements from smallest to largest atomic size:

I. barium	II. strontium	III. bismuth	IV. iodine
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 - III, I, II, IV
 - IV, III, II, I
 - IV, II, III, I
 - IV, II, I, III
- Which shape and bond angle is predicted by VSEPR theory for NF_3 ?
 - trigonal pyramid, less than 109.5°
 - tetrahedral, 109.5°
 - trigonal planar, 109.5°
 - trigonal planar, 120°

Answers To Unit 5 Post Test Chemistry

Marion Castellucci



Answers To Unit 5 Post Test Chemistry:

General Chemistry Darrell D. Ebbing, Larry K. Krannich, Joan I. Senyk, 1996 General Chemistry Study Guide Sixth Edition Darrell D. Ebbing, 1999 *Study Guide* Joan I. Senyk, Darrell D. Ebbing, 1984 **National Library of Medicine Audiovisuals Catalog** National Library of Medicine (U.S.), **Foundations Science** Janice Harriet Weiss, Contemporary, 1993-04 Contemporary s Foundation series provides thorough coverage of basic skills at reading levels 4-6 Gives students meaningful contexts for learning Makes materials easy to understand Provides students with the opportunity to create essay answers and practice the steps of the writing process Post tests assess skills proficiency upon completing each book Evaluation Charts target and prescribe areas for needed practice Thorough coverage of the writing process analyzing the essay writing the essay and mechanics **South-western GED Exercises** Marion Castellucci, 1995 **AGS GED Exercises** Marion Castellucci, 1995 **Using Constructivism in Teaching AP Chemistry** Avideh Lotfi, 2004 **Building Basic Skills in Science** Contemporary Books, inc, 1981 Contemporary s seven book series is a comprehensive instructional program for adult learners and pre high school skill levels The series develops solid skill foundations needed by learners who may go on to prepare for the high school equivalency examination Building Basic Skills is ideal for individualized or group instruction Each worktext has clear simple directions and easy to follow skill progressions for the learner working in a self paced program Each book provides everything the learner needs to master the basic skills *Development of a Self-instructional Unit on the Management of End Stage Renal Disease* Diane Gay Groves, 1977 **Solving Science Problems** Wilbur Lee Beauchamp, 1957 Polarity, Solutions, and Separation Science Kenda Jo Lemont, 1998 A Constructivist Approach to the Study of Matter Susan Hankins Fritzell, 2000 **Journal of Gerontological Nursing** , 1992 Science Problems for Junior High School Wilbur Lee Beauchamp, John Cunliffe Mayfield, Joe Young West, 1953 **The Effect of Adding Guided-inquiry to Laboratory Activities in an Acid Base Unit in a High School Chemistry Classroom** Kendon Douglas Smith, 2010 **Otto E. Miller, Plaintiff-Respondent, Against Fred W. Smythe, Defendant-Appellant** , Clinical Chemistry , 1968 **Manual of Chemical Engineering Shortcuts and Calculations** Chemical Engineering, 1953 **Automated Education Handbook** , 1969

This book delves into Answers To Unit 5 Post Test Chemistry. Answers To Unit 5 Post Test Chemistry is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Answers To Unit 5 Post Test Chemistry, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:
 - Chapter 1: Introduction to Answers To Unit 5 Post Test Chemistry
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 - Chapter 3: Answers To Unit 5 Post Test Chemistry in Everyday Life
 - Chapter 4: Answers To Unit 5 Post Test Chemistry in Specific Contexts
 - Chapter 5: Conclusion
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web counselling skills training in adolescent sexuality and reproductive health page 12 about this guide this guide is designed to help facilitators to run a 5 day workshop for training counselling skills in adolescent sexuality and reproductive health it is divided into six sections the first describes preparatory activities and the facilities