

Unit Test 1: Matter, Chemical Trends, & Chemical Bonding**Multiple Choice (10 marks) (KU: 3, C: 3, I: 2)***Identify the choice that best completes the statement or answers the question. Place your answers in the key below.*

1. ____ 2. ____ 3. ____ 4. ____ 5. ____

6. ____ 7. ____ 8. ____ 9. ____ 10. ____

1. The element found in the periodic table in Group 6 and Period 5 is
 - a. tungsten, W
 - b. molybdenum, Mo
 - c. tantalum, Ta
 - d. tellurium, Te
2. Which of the following groups is known for having both metallic and non-metallic properties?
 - a. alkali metals
 - b. alkali earth metals
 - c. transition metals
 - d. metalloids
3. How many electrons does a strontium atom have in its outer shell?
 - a. 1
 - b. 2
 - c. 7
 - d. 8
4. Which particles are responsible for the similar chemical and physical properties of elements in the same group?
 - a. protons and neutrons
 - b. number of energy shells
 - c. valence electrons
 - d. isotopes
5. Which of the following chemical formulas most likely represents a pure substance containing ionic bonds?
 - a. Cl_2
 - b. SO_3
 - c. CO_2
 - d. BaO
6. Consider the equation $\text{X}_{(g)} + e^- \rightarrow \text{X}^- + \text{energy}$. The "energy" term in the equation represents
 - a. electronegativity
 - b. electron affinity
 - c. ionization energy
 - d. nuclear charge
7. Elements A, B, C, and D (found in Groups 1-17) have atomic radii of $265 \text{ }\mu\text{m}$, $160 \text{ }\mu\text{m}$, $185 \text{ }\mu\text{m}$, and $175 \text{ }\mu\text{m}$, respectively. Which element will most likely have the lowest ionization energy?
 - a. A
 - b. B
 - c. C
 - d. D
8. List the following elements from smallest to largest atomic size:
 - I. barium
 - II. strontium
 - III. bismuth
 - IV. iodine
 - a. III, I, II, IV
 - b. IV, III, II, I
 - c. IV, III, III, I
 - d. IV, III, I, III
9. Which shape and bond angle is predicted by VSEPR theory for NF_3 ?
 - a. trigonal pyramidal, less than 109.5°
 - b. tetrahedral, 109.5°
 - c. trigonal planar, 109.5°
 - d. trigonal planar, 120°

Answers To Unit 5 Post Test Chemistry

Marion Castellucci

Answers To Unit 5 Post Test Chemistry:

General Chemistry Darrell D. Ebbing,Larry K. Krannich,Joan I. Senyk,1996 General Chemistry Study Guide Sixth Edition Darrell D. Ebbing,1999 Study Guide Joan I. Senyk,Darrell D. Ebbing,1984 **National Library of Medicine Audiovisuals Catalog** National Library of Medicine (U.S.), **Foundations Science** Janice Harriet Weiss,Contemporary,1993-04 Contemporary s Foundation series provides thorough coverage of basic skills at reading levels 4-6 Gives students meaningful contexts for learning Makes materials easy to understand Provides students with the opportunity to create essay answers and practice the steps of the writing process Post tests assess skills proficiency upon completing each books Evaluation Charts target and prescribe areas for needed practice Thorough coverage of the writing process analyzing the essay writing the essay and mechanics **South-western GED Exercises** Marion Castellucci,1995

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This book delves into Answers To Unit 5 Post Test Chemistry. Answers To Unit 5 Post Test Chemistry is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Answers To Unit 5 Post Test Chemistry, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Answers To Unit 5 Post Test Chemistry
- Chapter 2: Essential Elements of Answers To Unit 5 Post Test Chemistry
- Chapter 3: Answers To Unit 5 Post Test Chemistry in Everyday Life
- Chapter 4: Answers To Unit 5 Post Test Chemistry in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Answers To Unit 5 Post Test Chemistry. The first chapter will explore what Answers To Unit 5 Post Test Chemistry is, why Answers To Unit 5 Post Test Chemistry is vital, and how to effectively learn about Answers To Unit 5 Post Test Chemistry.
3. In chapter 2, the author will delve into the foundational concepts of Answers To Unit 5 Post Test Chemistry. The second chapter will elucidate the essential principles that need to be understood to grasp Answers To Unit 5 Post Test Chemistry in its entirety.
4. In chapter 3, this book will examine the practical applications of Answers To Unit 5 Post Test Chemistry in daily life. This chapter will showcase real-world examples of how Answers To Unit 5 Post Test Chemistry can be effectively utilized in everyday scenarios.
5. In chapter 4, this book will scrutinize the relevance of Answers To Unit 5 Post Test Chemistry in specific contexts. The fourth chapter will explore how Answers To Unit 5 Post Test Chemistry is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, the author will draw a conclusion about Answers To Unit 5 Post Test Chemistry. The final chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly recommended for anyone seeking to gain a comprehensive understanding of Answers To Unit 5 Post Test Chemistry.

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web counselling skills training in adolescent sexuality and reproductive health page 12 about this guide this guide is designed to help facilitators to run a 5 day workshop for training counselling skills in adolescent sexuality and reproductive health it is divided into six sections the first describes preparatory activities and the facilities