

## 19.2 Hydrogen Ions and Acidity

- The reaction in which water molecules produce ions is called the self-ionization of water
- $\text{H}_2\text{O(l)} \rightleftharpoons \text{H}^+(\text{aq}) + \text{OH}^-(\text{aq})$
- $2\text{H}_2\text{O(l)} \rightleftharpoons \text{H}^3\text{O}^+(\text{aq}) + \text{OH}^-(\text{aq})$



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**Guillaume Favre**

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**Physical Chemistry for Engineering and Applied Sciences** Frank R. Foulkes, 2012-09-12 Physical Chemistry for Engineering and Applied Sciences is the product of over 30 years of teaching first year Physical Chemistry as part of the Faculty of Applied Science and Engineering at the University of Toronto. Designed to be as rigorous as compatible with a first year student's ability to understand the text presents detailed step by step Memmler's Structure & Function of the Human Body, Enhanced Edition Barbara Janson Cohen, Kerry L. Hull, 2020-08-03 Continuing the tradition of excellence that has made it the preferred A P resource for allied health students the latest edition of Memmler's Structure and Function of the Human Body prepares you for success in your healthcare careers through easy to understand beautifully illustrated coverage of **Carbonic Acid Compounds and Hydrogen Ion Activities in Blood and Salt Solutions** Erik Johan Warburg, 1922 **Acidic Deposition: Terrestrial, materials, health and visibility effects** Patricia M. Irving, 1991

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**Medicus**, 1991 The Fundamentals of College Chemistry George Brooks King, William Elmer Caldwell, 1959 **Paper**  
Trade Journal, 1938-07 Technical Bulletin, 1923 Collected Papers Lister Institute of Preventive Medicine, 1912

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