

4 Atomic Structure



ELECTRONS AND THE STRUCTURE OF ATOMS

4.1 Defining the Atom



For students using the Foundation edition,
assign problems 1, 3-7.

Essential Understanding Atoms are the fundamental building blocks of matter.

Lesson Summary

Early Models of the Atom The scientific study of the atom began with John Dalton in the early 1800s.

- The ancient Greek Democritus first proposed that matter is made up of small, indivisible particles that he called atoms.
- John Dalton made the first accepted theory on atoms almost 2000 years after the work of Democritus.
- Dalton's atomic theory included that all atoms of an element are alike, the atoms of different elements are different, and atoms can combine to form compounds.

Sizing up the Atom Atoms are extremely small, but technology enables scientists to view atoms.

- An atom is the smallest part of an element that has the properties of that element.
- Individual atoms can be seen and even moved around using instruments such as scanning electron microscopes.

After reading Lesson 4.1, answer the following questions.

Early Models of the Atom

1. Democritus, who lived in Greece during the fourth century B.C., suggested that matter is made up of tiny particles that cannot be divided. He called these particles atoms.
2. List two reasons why the ideas of Democritus were not useful in a scientific sense.
They did not explain chemical behavior, and they lacked experimental support because Democritus's approach was not based on the scientific method.
3. The modern process of discovery about atoms began with the theories of an English schoolteacher named John Dalton.

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John Ogilvie

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1. Understanding the eBook 6 Self Check Activity Answers Pearson
 - The Rise of Digital Reading 6 Self Check Activity Answers Pearson
 - Advantages of eBooks Over Traditional Books
2. Identifying 6 Self Check Activity Answers Pearson
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 6 Self Check Activity Answers Pearson
 - User-Friendly Interface
4. Exploring eBook Recommendations from 6 Self Check Activity Answers Pearson
 - Personalized Recommendations
 - 6 Self Check Activity Answers Pearson User Reviews and Ratings
 - 6 Self Check Activity Answers Pearson and Bestseller Lists
5. Accessing 6 Self Check Activity Answers Pearson Free and Paid eBooks
 - 6 Self Check Activity Answers Pearson Public Domain eBooks
 - 6 Self Check Activity Answers Pearson eBook Subscription Services
 - 6 Self Check Activity Answers Pearson Budget-Friendly Options

6. Navigating 6 Self Check Activity Answers Pearson eBook Formats
 - ePUB, PDF, MOBI, and More
 - 6 Self Check Activity Answers Pearson Compatibility with Devices
 - 6 Self Check Activity Answers Pearson Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 6 Self Check Activity Answers Pearson
 - Highlighting and Note-Taking 6 Self Check Activity Answers Pearson
 - Interactive Elements 6 Self Check Activity Answers Pearson
8. Staying Engaged with 6 Self Check Activity Answers Pearson
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 6 Self Check Activity Answers Pearson
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 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 6 Self Check Activity Answers Pearson
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 6 Self Check Activity Answers Pearson
 - Setting Reading Goals 6 Self Check Activity Answers Pearson
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 6 Self Check Activity Answers Pearson
 - Fact-Checking eBook Content of 6 Self Check Activity Answers Pearson
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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