

## THERMODYNAMICS

## EXERCISE – I

## SINGLE CORRECT (OBJECTIVE QUESTIONS)

Sol.1 [C]

Sol.2 [D]

For 3 moles of  $\text{Cl}_2$  &  $\text{HCl}$   
 $\Delta U = -3 \times 185 = -555 \text{ kJ}$

Sol.3 [D]

2 mole  $\text{HCl}$  is consumed completely  
 for 1 mole consumption of  $\text{HCl}$   
 $\Delta U = \Delta H - \Delta n_g RT$

$$= -72.3 + \frac{8.314 \times 300}{1000} \quad (\text{as, } \Delta n_g = -1)$$

$$= -69.8 \text{ kJ}$$

For 2 mole  $\text{HCl}$ 

$$\Delta U = 2x - 69.8 = -139.6 \text{ kJ}$$

Sol.4 [B]

$$\Delta U = q + w$$

$$= \frac{100 \times 10^3 - 1.5 \times 10^5}{10^3} = -50 \text{ kJ}$$

Sol.5 [C]

$$w = nC_v \Delta T$$

$$\Rightarrow -3 \times 1000 = 1 \times 20 \times (T - 300)$$

$$T = 150 \text{ K}$$

Sol.6

$$\Delta U = Q + w$$

$$Q = -124 \text{ J}$$

$$W = -P_{\text{ext}}(V_2 - V_1)$$

$$= -\frac{1520}{760} (177 - 377) = 400 \text{ Atm ml} = 40 \text{ J}$$

$$\Delta U = -124 + 40 = -84 \text{ J}$$

Sol.7 [C]

Heat released by 500 gm of liquid

$$= \frac{500}{18} \times 75.6 \times 20$$

Heat absorbed by 'n' no of ice each 9 gm

$$= \frac{9}{12} n \times 6000 \text{ J}$$

Heat absorbed = Heat released

Which gives  $n = 14$ 

Sol.8 [A]

$$\text{Area of } \triangle ABC = -\text{Area of } \triangle ACD$$

Sol.9 [B]

$$C_p = \frac{1}{n} \left( \frac{dq}{dT} \right)_p$$

$$dT = 0$$

Sol.10 [C]

Sol.11 [D]

$$nC_v \Delta T = 100 \times 4.2 - 209$$

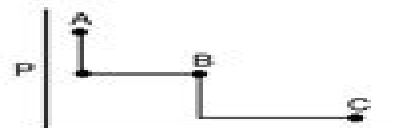
$$= 420 - 209 = 211 \text{ J}$$

$$n \times \frac{5R}{2} \times \Delta T = 211$$

$$n\Delta T = \frac{211 \times 2}{5R}$$

$$C_m = \frac{q}{n\Delta T} = \frac{420}{211 \times 2} \times 5R = 5R$$

Sol.12 [C]



$$\text{Total work done} = -2(16 - 4) - 1(32 - 16)$$

$$= -24 - 16 = -40 \text{ bar litre}$$

$$= -4000 \text{ J}$$

$$q = -w = 4000 \text{ J}$$

Sol.13 [A]

Sol.14 [C]

$$\gamma = \frac{7}{5}$$

$$\gamma - 1 = \frac{2}{5}$$

$$T_2 = T_1 \left( \frac{V_1}{V_2} \right)^{\frac{2}{5}}$$

$$= 600 \times \left( \frac{1}{32} \right)^{2/5} = 150 \text{ K}$$

$$\therefore \Delta H = nC_p (T_2 - T_1) = -\frac{7}{2} R \times 450 = -1575 R$$

# Answer Key Thermodynamics Physics

**Ebenezer Coloham Brewer**



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*Heat and the Principles of Thermodynamics* Charles Henry Draper, 1906

**Commonly Asked Questions**

**in Thermodynamics** Marc J. Assael, William A. Wakeham, Anthony R. H. Goodwin, Stefan Will, Michael Stamatoudis, 2011-03-10 Have you ever had a question that keeps persisting and for which you cannot find a clear answer Is the question seemingly so simple that the problem is glossed over in most resources or skipped entirely CRC Press Taylor and Francis is pleased to introduce Commonly Asked Questions in Thermodynamics the first in a new series of books that address the questions that frequently arise in today's major scientific and technical disciplines Designed for a wide audience from students and researchers to practicing professionals in related areas the books are organized in a user friendly Question Answer format Presented questions become increasingly specific throughout the book with clear and concise answers as well as illustrations diagrams and tables are incorporated wherever helpful Thermodynamics is a core discipline associated with the theoretical principles and practical applications underlying almost every area of science from nanoscale biochemical engineering to astrophysics Highlighting chemical thermodynamics in particular this book is written in an easy to understand style and provides a wealth of fundamental information simple illustrations and extensive references for further research and collection of specific data Designed for an audience that ranges from undergraduate students to scientists and engineers at the forefront of research this indispensable guide presents clear explanations for topics with wide applicability It reflects the fact that very often the most common questions are also the most profound *The Old Riddle and the Newest Answer* John Gerard, 1904 **Elementary treatise on physics, experimental and applied, tr. and ed. by E. Atkinson** Adolphe Ganot, 1893 **A treatise on arithmetic. [With] Key** James Thomson, 1881 **Physics, Uspekhi**, 1998 *Super Course for the MCAT* Stefan Bosworth, 1992-03 This brand new addition to Arco's acclaimed SuperCourse series is certain to set a new high standard for quality It offers the comprehensive MCAT preparation including in depth subject reviews as well as full length sample exams and in depth preparation for the new required essay section *Books and Pamphlets, Including Serials and Contributions to Periodicals* Library of Congress. Copyright Office, 1977 *The Bookseller*, 1894 *The Public School Latin Primer* Benjamin Hall Kennedy, 1878 **A Guide to English Composition** Ebenezer Coloham Brewer, 1878 *The Bookseller and the Stationery Trades' Journal*, 1894 Official organ of the book trade of the United Kingdom *Proceedings of the American Association for the Advancement of Science* American Association for the Advancement of Science, 1883 *Proceedings of the American Association for the Advancement of Science*, 1883 **Journal of Non-equilibrium Thermodynamics**, 1987 **NII Today**, 2013 *The Publishers' Trade List Annual*, 1980 *Thermal Physics* Ralph Baierlein, 1999 This textbook provides a clear instructive and highly readable introduction to thermal physics

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