

2

MATTER AND CHANGE

SECTION 2.1 PROPERTIES OF MATTER (pages 39–42)

This section helps you distinguish extensive from intensive properties and identify substances by their properties. It teaches you how to differentiate the three states of matter. It also defines a physical property and lists examples of physical properties and physical changes.

► Describing Matter (page 39)

1. The _____ mass _____ of an object is a measure of the amount of matter the object contains.

2. How does an extensive property differ from an intensive property?

An extensive property depends on the amount of matter, while an intensive property depends on the type of matter.

► Identifying Substances (page 40)

3. Matter that has a uniform and definite composition is called a _____ substance _____.

4. Is the following sentence true or false? All samples of a substance have different physical properties. _____ false _____

5. A physical property is a quality or condition of a substance that can be _____ observed _____ or _____ measured _____ without changing the substance's composition.

6. Circle the letter of the term that is NOT a physical property.

- a. hardness c. boiling point
b. color **d. melting**

7. Look at Table 2.1 on page 40. What is the melting point of bromine? _____ -7°C _____

8. Look at Table 2.1 on page 40. Circle the letter of the substance that is a yellow solid and melts at 115°C.

- a. sulfur**
b. chlorine
c. gold
d. copper

Section 3 1 Chemistry Answers

ML Morrison



Section 3 1 Chemistry Answers:

Basic Concepts of Chemistry Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section and relevant chapter problems at the end of each chapter. Every concept in the text is clearly illustrated with one or more step by step examples. Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning. This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. WileyPLUS sold separately from text.

GCSE Additional Science Higher Hannah Kingston, 2006-09 Covers the 2006 Gateway Additional Science specification for all exam boards AQA Edexcel and OCR for students going on to study Additional Science. Part of the Success series, this title emphasises the shift from fact learning to investigating and understanding how science works.

The student's handbook of the practice of medicine Henry Aubrey Husband, 1900 **Russian Journal of General Chemistry**, 1998 Study Guide and Partial Solutions Manual, Fundamentals of General, Organic, and Biological Chemistry Susan McMurry, 1996 This internationally acclaimed detective series is just the thing for lovers of those Number One Ladies looking for a darker more realistic view of Botswana. Sue Baker Publishing News A classified catalogue of ... educational works in use in Great Britain [by W. Low]. Walter Low, 1871 *Journal of General Chemistry of the U.S.S.R. in English Translation*, 1971 **Transactions** Homœopathic Medical Society of the State of New York, 1890 List of members in each vol Academy, with which are Incorporated Literature and the English Review, 1891 *The Saturday Review of Politics, Literature, Science and Art*, 1892 The Academy, 1891 **The Chemical News and Journal of Physical Science**, 1886 **The Lancet**, 1888 Chemistry J. J. Thompson, 1972 **Journal of the Chemical Society**, 1989 **The Athenaeum**, 1874 **The National Druggist**, 1910 **The Naturalists' Leisure Hour and Monthly Bulletin**, 1881 **Journal of Chemical Education**, 1929 Chemical Engineering Catalog, 1922

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