

Activity A: Subatomic particles	Get the Gizmo ready: <ul style="list-style-type: none"> Use the arrows to create an atom with two protons, two neutrons, and two electrons. Turn on Show element name. 	
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Question: What are the properties of protons, neutrons, and electrons?

1. **Observe:** Turn on Show element symbol and Element notation. Three numbers surround the element symbol: the **mass number** (A), electrical charge (no number is displayed if the atom is neutral), and the **atomic number** (Z).
2. **Investigate:** Watch how the numbers change as you add or remove particles.



- Which number is equal to the number of protons in the atom?
- How can you calculate the number of neutrons (N) in an atom?
- Which particle (proton, neutron, or electron) has a positive charge?
- Negative charge?

neutrons
 by the number of protons

electron
 proton

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Holt Rinehart & Winston

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