

Test Bank for Chemistry 11th Edition Chang

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1. In a cathode ray tube
A) electrons pass from the anode to the cathode.
B) electrons pass from the cathode to the anode.
C) protons pass from the anode to the cathode.
D) protons pass from the cathode to the anode.
Ans: B Category: Medium Section: 2.2
2. The scientist who determined the magnitude of the electric charge of the electron was
A) John Dalton D) Henry Moseley
B) Robert Millikan E) R. Chang
C) J. J. Thomson
Ans: B Category: Easy Section: 2.2
3. When J. J. Thomson discovered the electron, what physical property of the electron did he measure?
A) its charge, e D) its mass, m
B) its charge-to-mass ratio, e/m E) its atomic number, Z
C) its temperature, T
Ans: B Category: Easy Section: 2.2
4. Which of the following scientists developed the nuclear model of the atom?
A) John Dalton D) Henry Moseley
B) Robert Millikan E) Ernest Rutherford
C) J. J. Thomson
Ans: E Category: Easy Section: 2.2
5. Rutherford's experiment with alpha particle scattering by gold foil established that
A) protons are not evenly distributed throughout an atom.
B) electrons have a negative charge.
C) electrons have a positive charge.
D) atoms are made of protons, neutrons, and electrons.
E) protons are 1840 times heavier than electrons.
Ans: A Category: Medium Section: 2.2
6. Atoms of the same element with different mass numbers are called
A) ions. B) neutrons. C) allotropes. D) chemical families. E) isotopes.
Ans: E Category: Easy Section: 2.3

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Lei Shi



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1. Understanding the eBook Answer Key To Chemistry 11th Edition Chang
 - The Rise of Digital Reading Answer Key To Chemistry 11th Edition Chang
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key To Chemistry 11th Edition Chang
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key To Chemistry 11th Edition Chang
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key To Chemistry 11th Edition Chang
 - Personalized Recommendations
 - Answer Key To Chemistry 11th Edition Chang User Reviews and Ratings
 - Answer Key To Chemistry 11th Edition Chang and Bestseller Lists
5. Accessing Answer Key To Chemistry 11th Edition Chang Free and Paid eBooks

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 - Answer Key To Chemistry 11th Edition Chang Enhanced eBook Features
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 - Highlighting and Note-Taking Answer Key To Chemistry 11th Edition Chang
 - Interactive Elements Answer Key To Chemistry 11th Edition Chang
8. Staying Engaged with Answer Key To Chemistry 11th Edition Chang
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key To Chemistry 11th Edition Chang
9. Balancing eBooks and Physical Books Answer Key To Chemistry 11th Edition Chang
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key To Chemistry 11th Edition Chang
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Answer Key To Chemistry 11th Edition Chang
- Setting Reading Goals Answer Key To Chemistry 11th Edition Chang
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Key To Chemistry 11th Edition Chang
- Fact-Checking eBook Content of Answer Key To Chemistry 11th Edition Chang
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

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- Interactive and Gamified eBooks

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