

Chapter 10 Nuclear Chemistry

Section 10.3 Artificial Transmutation**[pages 363–366]**

This section discusses transmutations, transuranium elements, and particle accelerators.

Reading Strategy [page 363]

Monitoring Your Understanding: Preview the Key Concepts, topic headings, vocabulary, and figures in this section. List two things you expect to learn. After reading, state what you learned about each item you listed. For more information on this Reading Strategy, see the Reading and Study Skills in the Skills and Reference Handbook at the end of your textbook.

Understanding Artificial Transmutation	
What I Expect to Learn	What I Learned

Nuclear Reactions in the Laboratory [page 363]

1. Define transmutation. _____
2. An example of a transmutation that occurs naturally is _____. _____
3. How do scientists perform artificial transmutations? _____
4. Circle the letter that identifies the scientist who performed the first artificial transmutation.

a. Ernest Rutherford	b. Niels Bohr
c. Enrico Fermi	d. Luis Alvarez
5. The experiment that produced the first artificial transmutation also provided evidence that the nucleus contains _____. _____

Transuranium Elements [page 364]

6. Describe a transuranium element. _____
7. Is the following sentence true or false? All transuranium elements are radioactive. _____

Answer Key For Section 10 3 Artificial Transmutation

Rolland J. Gladieux

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IIT Chemistry-I, **The Pearson Guide To Physical Chemistry**

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