



13. Complete the following table for the three experiments in Model 2.

	Experiment C	Experiment D	Experiment E
Independent Variable	# of particles	Temp.	Internal External Pressure
Dependent Variable	Volume	Volume	Volume
Controlled Variable(s)	Temp / Internal & External Pressure / # of particles	Temp / # of particles	

14. Provide a molecular level explanation for the increase in volume among the balloons in Experiment C. (How often and/or how hard are the molecules hitting the sides of the container?)

of molecules ↑, so number of collisions ↑ + Volume along with it
collisions ↑ energy stays the same

15. Provide a molecular level explanation for the increase in volume among the balloons in Experiment D.

Temperature ↑ increasing the kinetic energy of the particles as well
which ↑ collisions thus ↑ pressure

16. Provide a molecular level explanation for the increase in volume among the balloons in Experiment E.

if external forces decrease then it allows internal forces ↑
thus ↑ volume

Answer Key For Pogil Gas Variables

David Kirk

Answer Key For Pogil Gas Variables:

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In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has been more apparent than ever. Its capability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Answer Key For Pogil Gas Variables**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we will delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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