

# Compounds

**BEFORE YOU READ**

After you read this section, you should be able to answer these questions:

- What are compounds?
- What is a chemical reaction?
- How are compounds used in everyday life?

National Science  
Education Standards  
PS 1a, 1c

## What Are Compounds?

Look around the classroom. Most of the things you see are not made of just one element. Instead, they are made of elements combined with other elements. A **compound** is a pure substance composed of two or more elements that are joined by chemical bonds. The figure below shows some compounds that you might find in your kitchen and what elements make up the compounds. ☐

**Familiar Compounds**

Compound	Elements in the compound
Table salt	sodium and chlorine
Water	hydrogen and oxygen
Sugar	hydrogen, carbon, and oxygen
Carbon dioxide	carbon and oxygen
Baking soda	sodium, hydrogen, carbon, and oxygen

**RATIO OF ELEMENTS IN A COMPOUND**

Elements join in a specific ratio according to their masses to form a compound. For example, in 18 g of water, there are 2 g of hydrogen and 16 g of oxygen. The mass ratio of oxygen to hydrogen is  $\frac{2\text{ g}}{16\text{ g}}$  or  $\frac{1}{8}$ . The mass ratio is written as 1 to 8 or 1:8. Every sample of water has a 1:8 mass ratio of hydrogen to oxygen. What happens if a compound has a different mass ratio of hydrogen to water? The compound cannot be water.

Sometimes the same two elements can join in different ratios. However, two different compounds are formed. For example, carbon and oxygen can join to form carbon monoxide,  $\text{CO}$ , and can also form carbon dioxide,  $\text{CO}_2$ . Carbon dioxide forms when there is lots of oxygen present.

 **STUDY TIP**

**Asking Questions** Read this section silently. In your notebook, write down questions that you have about this section. Discuss your questions in a small group.

 **READING CHECK**

1. Describe What is a compound?

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**Math Focus**

2. Determine A compound has 40 g of calcium and 160 g of bromine. What is the mass ratio of calcium to bromine in the compound?

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# Answer Key For Holt Science Chemical Compounds

**C Cleary**

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## **Table of Contents Answer Key For Holt Science Chemical Compounds**

1. Understanding the eBook Answer Key For Holt Science Chemical Compounds
  - The Rise of Digital Reading Answer Key For Holt Science Chemical Compounds
  - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key For Holt Science Chemical Compounds
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Answer Key For Holt Science Chemical Compounds
  - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key For Holt Science Chemical Compounds
  - Personalized Recommendations
  - Answer Key For Holt Science Chemical Compounds User Reviews and Ratings
  - Answer Key For Holt Science Chemical Compounds and Bestseller Lists

5. Accessing Answer Key For Holt Science Chemical Compounds Free and Paid eBooks
  - Answer Key For Holt Science Chemical Compounds Public Domain eBooks
  - Answer Key For Holt Science Chemical Compounds eBook Subscription Services
  - Answer Key For Holt Science Chemical Compounds Budget-Friendly Options
6. Navigating Answer Key For Holt Science Chemical Compounds eBook Formats
  - ePUB, PDF, MOBI, and More
  - Answer Key For Holt Science Chemical Compounds Compatibility with Devices
  - Answer Key For Holt Science Chemical Compounds Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Answer Key For Holt Science Chemical Compounds
  - Highlighting and Note-Taking Answer Key For Holt Science Chemical Compounds
  - Interactive Elements Answer Key For Holt Science Chemical Compounds
8. Staying Engaged with Answer Key For Holt Science Chemical Compounds
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Answer Key For Holt Science Chemical Compounds
9. Balancing eBooks and Physical Books Answer Key For Holt Science Chemical Compounds
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Answer Key For Holt Science Chemical Compounds
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Answer Key For Holt Science Chemical Compounds
  - Setting Reading Goals Answer Key For Holt Science Chemical Compounds
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Answer Key For Holt Science Chemical Compounds
  - Fact-Checking eBook Content of Answer Key For Holt Science Chemical Compounds
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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