

Half Life Worksheet – Extra Practice

Answers
Below

- 1) Fluorine-21 has a half life of approximately 5 seconds. What fraction of the original nuclei would remain after 1 minute?
- 2) Iodine-131 has a half life of 8 days. What fraction of the original sample would remain at the end of 32 days?
- 3) The half-life of chromium-51 is 28 days. If the sample contained 510 grams, how much chromium would remain after 56 days? How much would remain after 1 year? How much was present 168 days ago?
- 4) If 20.0 g of a radioactive isotope are present at 1:00 PM and 5.0 g remain at 2:00 PM, what is the half life of the isotope?
- 5) ~~The half life of Uranium-238 is 4.5 billion years and the age of earth is 4.5×10^9 years. What fraction of Uranium-238 that was present when Earth was formed still remains?~~
- 6) Chromium-48 decays. After 6 half-lives, what fraction of the original nuclei would remain?
- 7) The half life of Iodine-125 is 60 days. What fraction of Iodine-125 nuclides would be left after 360 days?
- 8) Titanium-51 decays with a half life of 6 minutes. What fraction of titanium would remain after one hour?
- 9) ~~A medical institution requests 1 g of bismuth-214, which has a half life of 20 min. How many grams of bismuth-214 must be prepared if the shipping time is 2 h?~~
- 10) ~~The half life of radium 226 is 1602 years. If you have 500 grams of radium today how many grams would have been present 9612 years ago?~~

$$\textcircled{1} \quad \frac{1}{2^{n/2}} = \frac{1}{4096}$$

$$\textcircled{2} \quad \left(\frac{1}{2}\right)^7 = \frac{1}{128}$$

$$\textcircled{3} \quad 56 \text{ days} \\ 56 = 28(2) \\ 510 \cdot (0.5)^2 = 127.5 \text{ g}$$

$$\textcircled{4} \quad \#3 \quad 1 \text{ year} \\ (365 \text{ day})$$

$$510 \cdot (0.5) = 0.06 \text{ g}$$

1	0	20	60	min
2	20	10	5	
3	10	5	2.5	
4	5	2.5	1.25	

$$\textcircled{5} \quad \frac{1}{2^{n/6}} = \frac{1}{64}$$

$$\textcircled{6} \quad \frac{1}{2^{n/6}} = \frac{1}{64} \\ 360 \div 60 = 6$$

$$\textcircled{7} \quad 60 \div 6 = 10 \\ \frac{1}{2^{n/10}} = \frac{1}{1024}$$

Answer Key For Half Life Problems

Christopher John Willis

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Table of Contents Answer Key For Half Life Problems

1. Understanding the eBook Answer Key For Half Life Problems
 - The Rise of Digital Reading Answer Key For Half Life Problems
 - Advantages of eBooks Over Traditional Books
2. Identifying Answer Key For Half Life Problems
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Answer Key For Half Life Problems
 - User-Friendly Interface
4. Exploring eBook Recommendations from Answer Key For Half Life Problems
 - Personalized Recommendations
 - Answer Key For Half Life Problems User Reviews and Ratings

- Answer Key For Half Life Problems and Bestseller Lists
- 5. Accessing Answer Key For Half Life Problems Free and Paid eBooks
 - Answer Key For Half Life Problems Public Domain eBooks
 - Answer Key For Half Life Problems eBook Subscription Services
 - Answer Key For Half Life Problems Budget-Friendly Options
- 6. Navigating Answer Key For Half Life Problems eBook Formats
 - ePUB, PDF, MOBI, and More
 - Answer Key For Half Life Problems Compatibility with Devices
 - Answer Key For Half Life Problems Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Answer Key For Half Life Problems
 - Highlighting and Note-Taking Answer Key For Half Life Problems
 - Interactive Elements Answer Key For Half Life Problems
- 8. Staying Engaged with Answer Key For Half Life Problems
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Answer Key For Half Life Problems
- 9. Balancing eBooks and Physical Books Answer Key For Half Life Problems
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Answer Key For Half Life Problems
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Answer Key For Half Life Problems
 - Setting Reading Goals Answer Key For Half Life Problems
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Answer Key For Half Life Problems
 - Fact-Checking eBook Content of Answer Key For Half Life Problems
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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