

CHEM107-Student Exploration: Nuclear Decay answer key-2024-2025

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

The chart below gives the locations, charges, and approximate masses of three subatomic particles. The approximate mass of each particle is given in universal mass units (u).

Particle	Location	Charge	Approximate mass
Proton	Nucleus	1^+	1 u
Neutron	Nucleus	0	1 u
Electron	Orbitals	1^-	0 u

1. The mass number of an atom is equal to the sum of protons and neutrons in the nucleus. A helium atom has 2 protons and 2 neutrons.

What is the mass number of this atom? 4

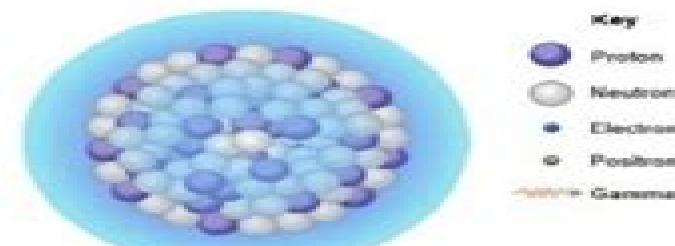
2. The atomic number of an element is equal to the number of protons in each atom of the element.

All helium atoms have 2 protons. What is the atomic number of helium? 2

Gizmo Warm-up

While most atoms are stable, some are radioactive, which means that they have a tendency to undergo spontaneous nuclear decay. The decay of radioactive atoms generally results in the emission of particles and/or energy.

Several types of nuclear decay can be explored with the *Nuclear Decay Gizmo*™. On the Gizmo, check that **Alpha decay** and **Uranium** are selected.



1. Click Play (▶), and then click Pause (⏸) when the alpha particle is clearly

Answer Key To Nuclear Decay

Xiaolong Qi

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