

Name _____

Date _____

Reading the Periodic Table KEY1. List the atomic numbers of the elements in Period 2.

3, 4, 5, 6, 7, 8, 9, 10

2. List the symbols of the elements in Period 2.

Na, Mg, Al, Si, P, S, Cl, Ar

3. Name the elements in Period 2 that are metals.

Li, Be

4. Name the elements in Period 1 that are nonmetals.

H, He

5. Which element in Period 2 has the least mass?

Sodium (Na)

6. Which element in Period 4 has the most mass?

Krypton (Kr)

7. To which group (family) do these belong?

a. Cobalt transition metals

d. Silicon group (I) nonmetals

b. Sodium alkali metals

e. Chlorine halogens

c. Argon noble gases

f. Magnesium alkaline earth metals

metals

8. Find iron on the periodic table:

a. What is the atomic number of iron? 26

b. To which group does iron belong? Transition metals

9. Name the elements that have properties similar to that of chlorine.

fluorine, bromine, iodine, chlorine, bromine

10. Fill in the chart below. The first row has been completed as an example.

Element	Symbol	Atomic Number	Period	Group
silicon	Si	14	3	14
nitrogen	N	7	2	15
krypton	Kr	36	4	18
gold	Au	79	6	19
chromium	Cr	24	4	18
chlorine	Cl	17	3	17

2 Exploring The Periodic Table Answer Key

Rachel E. Green

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