

Name: \_\_\_\_\_

Date: \_\_\_\_\_

### Reading the Periodic Table **KEY**

1. List the atomic numbers of the elements in Period 2.

**3, 4, 5, 6, 7, 8, 9, 10**

2. List the symbols of the elements in Period 3.

**Na, Mg, Al, Si, P, S, Cl, Ar**

3. Name the elements in Period 2 that are metals.

**Li, Be**

4. Name the elements in Period 1 that are nonmetals.

**H, He**

5. Which element in Period 3 has the least mass?

**Sodium (Na)**

6. Which element in Period 4 has the most mass?

**Krypton (Kr)**

7. To which group/family do these belong?

a. Cobalt **transition metals**

b. Sodium **alkali metals**

c. Argon **noble gases**

d. Silicon **group 14 metalloids**

e. Chlorine **halogens**

f. Magnesium **alkaline earth metals**

8. Find iron on the periodic table:

a. What is the atomic number of iron? **26**

b. To which group does iron belong? **Transition metals**

9. Name the elements that have properties similar to that of chlorine.

**Fluorine, bromine, iodine, astatine, tennessine**

10. Fill in the chart below. The first row has been completed as an example.

Element	Symbol	Atomic Number	Period	Group
sodium	Na	11	3	1
<b>oxygen</b>	<b>O</b>	<b>8</b>	<b>2</b>	<b>16</b>
krypton	Kr	36	4	18
gold	Au	79	6	11
cadmium	Cd	48	5	12
manganese	Mn	25	4	7

## 2 Exploring The Periodic Table Answer Key

**Rachel E. Green**



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web professor charles d kolstad of stanford university is an internationally known environmental economist with a focus on industrial organization and public economics his research interests are in information uncertainty and regulation with much of his applied work in the area of climate change and energy markets

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