

3 - 4 Review and Reinforcement

Elements and Compounds

Fill in each blank with the name of the element.

Elemental Matter	Elemental Compound	Elemental & Compound
1. Sodium	Na	Sodium
2. Magnesium	Mg	Magnesium
3. Carbon	C	Carbon
4. Hydrogen	H	Hydrogen
5. Iron	Fe	Iron
6. Zinc	Zn	Zinc
7. Potassium	K	Potassium
8. Copper	Cu	Copper
9. Silver	Ag	Silver
10. Nickel	Ni	Nickel

Complete the following by inserting the missing words appropriate to each blank line from the list:

element
compound
mixture

substance
mixture
element

physical matter
matter
chemical matter

1. Elements are matter composed of identical particles.
2. Elements are substances which are simple, stable and cannot be chemically decomposed.
3. Elements are substances which cannot be chemically decomposed.
4. Physical matter elements chemically combined make up compounds.
5. Chemistry is concerned with elements and compounds is the study of a compound.
6. Elements and compounds are called pure substances because they have a single composition and physical properties.

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Prentice-Hall Staff



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Matter, Building Block of the Universe ,1993 Matter Prentice-Hall Staff,1994 *Learning and Living in the 21st Century 6 Teacher's Manual 1st Ed. 2007* , Appleton's Review for National Boards Part II Robin J. O. Catlin,1985

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Reviewed in The Textbook Letter 3 4 94 *Parade of Life* PH Inc. Staff,1994 **Science** Herbert A. Smith,1966

Appleton and Lange's Review for the Physician Assistant Patrick J. Cafferty,1994 **Electricity and Magnetism** ,1993

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Philippine Journal of Education ,1976

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