

Name: KEY

Pr. _____

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Write balanced formula equations for these reactions and identify the type of reaction.

1. Potassium + Iodine	$2K + I_2 \rightarrow 2KI$	synthesis
2. Barium Iodide + Sodium Sulfate	$BaI_2 + Na_2SO_4 \rightarrow BaSO_4 + 2NaI$	double rep.
3. Aluminum Hydroxide	$2Al(OH)_3 \rightarrow Al_2O_3 + 3H_2O$	decomposition
4. Sodium Chlorate	$NaClO_3 \rightarrow NaCl + O_2$	decomposition
5. Aluminum Bromide + Chlorine	$AlBr_3 + Cl_2 \rightarrow AlCl_3 + Br_2$	single replacement
6. Phosphoric Acid + Magnesium	$H_3PO_4 + Mg \rightarrow Mg_3(PO_4)_2 + H_2$	single rep.
7. Iron III Sulfate + Calcium Nitrate	$Fe_2(SO_4)_3 + 3Ca(NO_3)_2 \rightarrow 2Fe(NO_3)_3 + 3CaSO_4$	double rep.
8. Sodium + Water	$2Na + 2H_2O \rightarrow 2NaOH + H_2$	single rep.
9. Aluminum + Iodine	$2Al + 3I_2 \rightarrow 2AlI_3$	synthesis
10. Iron III Hydroxide + Hydrobromic Acid	$Fe(OH)_3 + 3HBr \rightarrow FeBr_3 + 3H_2O$	double rep.
11. Magnesium Hydroxide + Ammonium Sulfate	$Mg(OH)_2 + (NH_4)_2SO_4 \rightarrow MgSO_4 + 2NH_4OH$	double rep.
12. Water	$2H_2O \xrightarrow{\text{electricity}} 2H_2 + O_2$	decomposition
13. Calcium + Hydrochloric Acid	$Ca + 2HCl \rightarrow CaCl_2 + H_2$	single replacement
14. Lithium + Chlorine	$2Li + Cl_2 \rightarrow 2LiCl$	synthesis

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Answer Key

Robert Mitchell

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Holt Chemistry Ralph Thomas Myers,2004 *Describing Chemical Engineering Systems* William Edwin Ranz,1970

General Chemistry Donald A. McQuarrie,Stanley Gill,2011-06-15 This Fourth Edition of McQuarrie's classic text offers a thorough revision and a quantum leap forward from the previous edition. Taking an atoms first approach it promises to be another ground breaking text in the tradition of McQuarrie's many previous works. This outstanding new text available in a soft cover edition offers professors a fresh choice and outstanding value. *Essentials of Chemical Reaction Engineering* H. Scott Fogler,2011 Accompanying DVD ROM contains many realistic interactive simulations **Chemical Interactions**

,2005 **Merrill Chemistry** Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 *Soil Chemistry* Daniel G. Strawn,Hinrich L. Bohn,George A. O'Connor,2020-01-28 Provides comprehensive coverage of the chemical interactions among organic and inorganic solids air water microorganisms and the plant roots in soil. This book focuses on the species and reaction processes of chemicals in soils with applications to environmental and agricultural issues. Topics range from discussion of fundamental chemical processes to review of properties and reactions of chemicals in the environment. This new edition contains more examples more illustrations more details of calculations and reorganized material within the chapters including nearly 100 new equations and 51 new figures. Each section also ends with an important concepts overview as well as new questions for readers to answer. Starting with an introduction to the subject *Soil Chemistry* 5th Edition offers in depth coverage of properties of elements and molecules characteristics of chemicals in soils soil water chemistry redox reactions in soils mineralogy and weathering processes in soils and chemistry of soil clays. The book also provides chapters that examine production and chemistry of soil organic matter surface properties of soil colloids adsorption processes in soils measuring and predicting sorption processes in soils soil acidity and salt affected soils. Provides a basic description of important research and fundamental knowledge in the field of soil chemistry. Contains more than 200 references provided in figure and table captions and at the end of the chapters. Extensively revised with updated figures and tables. *Soil Chemistry* 5th Edition is an excellent text for senior level soil chemistry students

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