

NFPA Symbol Explanation Guide

Health Hazard

4: Deadly
3: Extreme Danger
2: Hazardous
1: Slightly Hazardous
0: Normal Material

Fire Hazard

4: Below 73F
3: Below 100F
2: Below 200F
1: Above 200F
0: Will Not Burn

Specific Hazard

Oxidizer	OXY
Acid	ACID
Alkali	ALK
Corrosive	COR
Use NO WATER	W
Radiation Hazard	☢



Reactivity

4: May Detonate
3: Shock And Heat May Detonate
2: Violent Chemical Change
1: Unstable If Heated
0: Stable

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NFPA.



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web feb 25 2022 detailed experimental light brown nutmeg powder 4 g was mixed with 25 ml of clear and colorless diethyl
ether solvent and the brown cloudy mixture was filtered by gravity filtration with fluted fast flow filter paper in a powder
funnel into a 250 ml round bottom flask rbf leaving light brown residue and beige filtrate diethyl ether was

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web trimyristin in the major triglyceride found in nutmeg representing 20 25 of the dried weight of ground nutmeg to extract
the trimyristin we performed a solid liquid extraction using the organic solvent methylene chloride CH_2Cl_2

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web isolation of trimyristin from nutmeg purpose the purpose of this experiment is to isolate myristic acid from ground
nutmeg seeds nutmeg seeds are extracted using tbme to yield crude trimyristin then recrystallized to yield pure trimyristin
which will be hydrolyzed and undergo esterification

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web read standard reflux on page 201 of the organic chem lab survival manual prelab question 1 trimyristin is a triglyceride
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nutmeg record exact mass and 15 ml of diethyl ether to microwave vessel

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web trimyristin is found in the fixed oil of nutmeg the fixed oil comprises approximately 24 40 of the nutmeg seed trimyristin comprises 73 of the fixed oil overall trimyristin should have percent recovery of 18 29 1 figure 1 shows how trimyristin is triester formed from the dehydration reaction between glycerol and myristic acid oh

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web in this lab trimyristin is extracted from ground nutmeg occurring in many vegetable oils as well as myristic acid trimyristin consists of long saturated hydrocarbon chains and it is relatively nonpolar thus using a relatively nonpolar solvent trimyristin can be easily extracted by the concept of like dissolves like

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web the desired components of solids are dissolved in a solvent then the remaining solid is filtered so that the selected element and solvent are left over in this lab the trimyristin was dissolved into the solvent and the remainder of the nutmeg was filtered out

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