

# Atomic Structure



Atomic Number

Element Symbol

Element Name

Atomic Mass

The atomic number represents these two parts of the atom:  
number of protons & number of electrons

You can calculate the number of neutrons by subtracting

Atomic Mass minus Atomic Number



Protons:

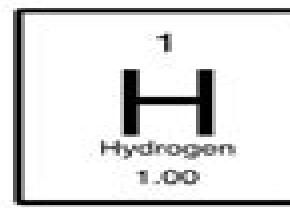
7

Electrons:

7

Neutrons:

7



Protons:

1

Electrons:

1

Neutrons:

0



Protons:

8

Electrons:

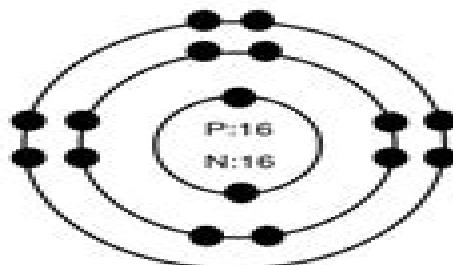
8

Neutrons:

8

## BOHR DIAGRAMS

- Draw a circle in the center representing the nucleus.
- Label the number of protons and neutrons.
- Draw the electrons in their energy levels.
  1. The first energy level can hold 2 electrons
  2. The second energy level can hold 8 electrons
  3. The third energy level can hold 8 electrons

Element: SulfurNumber of Valence Electrons: 6

The electrons in the outermost energy level are called:

Valence Electrons

## LEWIS DIAGRAMS

A Lewis diagram shows only the symbol with the Valence electrons



# Atomic Answer Key

**Rachel E. Green**

## Atomic Answer Key:

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## **Table of Contents Atomic Answer Key**

1. Understanding the eBook Atomic Answer Key
  - The Rise of Digital Reading Atomic Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying Atomic Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Atomic Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from Atomic Answer Key
  - Personalized Recommendations
  - Atomic Answer Key User Reviews and Ratings
  - Atomic Answer Key and Bestseller Lists
5. Accessing Atomic Answer Key Free and Paid eBooks
  - Atomic Answer Key Public Domain eBooks

- Atomic Answer Key eBook Subscription Services
- Atomic Answer Key Budget-Friendly Options

6. Navigating Atomic Answer Key eBook Formats

- ePub, PDF, MOBI, and More
- Atomic Answer Key Compatibility with Devices
- Atomic Answer Key Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of Atomic Answer Key
- Highlighting and Note-Taking Atomic Answer Key
- Interactive Elements Atomic Answer Key

8. Staying Engaged with Atomic Answer Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Atomic Answer Key

9. Balancing eBooks and Physical Books Atomic Answer Key

- Benefits of a Digital Library
- Creating a Diverse Reading Collection Atomic Answer Key

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine Atomic Answer Key

- Setting Reading Goals Atomic Answer Key
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of Atomic Answer Key

- Fact-Checking eBook Content of Atomic Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

#### 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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