

- (a) From the multiplication rule,  $10^3 = 1000$  prefixes are possible.  
 (b) From the multiplication rule,  $8 \times 2 \times 10 = 160$  are possible.  
 (c) Every arrangement of these digits selected from the 10 digits results in a possible prefix.  
 $P_3^{10} = \frac{10!}{7!} = 720$  prefixes are possible.

2-46. A byte is a sequence of eight bits and each bit is either 0 or 1.

- (a) How many different bytes are possible?  
 (b) If the first bit of a byte is a parity check, that is, the first byte is determined from the other seven bits, how many different bytes are possible?

- (a) From the multiplication rule,  $2^8 = 256$  bytes are possible.  
 (b) From the multiplication rule,  $2^7 = 128$  bytes are possible.

2-47. In a chemical plant, 24 holding tanks are used for final product storage. Four tanks are selected at random and without replacement. Suppose that six of the tanks contain material in which the viscosity exceeds the customer requirements.

- (a) What is the probability that exactly one tank in the sample contains high-viscosity material?  
 (b) What is the probability that at least one tank in the sample contains high-viscosity material?  
 (c) In addition to the six tanks with high viscosity levels, four different tanks contain material with high impurities. What is the probability that exactly one tank in the sample contains high-viscosity material and exactly one tank in the sample contains material with high impurities?

(a) The total number of samples possible is  $\binom{24}{4} = \frac{24!}{4!20!} = 10,626$ . The number of samples in which exactly one tank has high viscosity is  $\binom{6}{1} \binom{18}{3} = \frac{6!}{1!5!} \times \frac{18!}{3!15!} = 4896$ . Therefore, the probability is

$$\frac{4896}{10626} = 0.461$$

(b) The number of samples that contain no tank with high viscosity is  $\binom{18}{4} = \frac{18!}{4!14!} = 3060$ . Therefore, the requested probability is  $1 - \frac{3060}{10626} = 0.712$ .

(c) The number of samples that meet the requirements is  $\binom{6}{1} \binom{4}{1} \binom{18}{2} = \frac{6!}{1!5!} \times \frac{4!}{1!3!} \times \frac{18!}{2!16!} = 2184$ . Therefore, the probability is  $\frac{2184}{10626} = 0.206$ .

2-48. Plastic parts produced by an injection-molding operation are checked for conformance to specifications. Each tool contains 12 cavities in which parts are produced, and these parts fall into a conveyor when the press opens. An inspector chooses 3 parts from among the 12 at random. Two cavities are affected by a temperature malfunction that results in parts that do not conform to specifications.

- (a) How many samples contain exactly 1 nonconforming part?  
 (b) How many samples contain at least 1 nonconforming part?

# Applied Statistics And Probability For Engineers Answer Key

**Patricia Dominguez-Marin**

## **Applied Statistics And Probability For Engineers Answer Key:**

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### The Discrete Ordered Median Problem: Models and Solution Methods Patricia Dominguez-Marin, 2013-12-11

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for the nitration 4 01 ml of nitric acid 4 00 ml of sulfuric acid and 4 56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4 56 ml of bromobenzene was used the amount of product predicted to be recovered was 8 7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction eas particularly nitration of methyl benzoate go give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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section 20711 lab experiment nitration of bromobenzene introduction the chemical reaction of nitration consists of a nitro group being added to or substituted in a molecule nitration can basically be carried out by a mixture of concentrated nitric acid and sulfuric acid this mixture is useful to obtain the active nitronium ion

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical x 100 what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area theory and explanation of the nitration of bromobenzene if the temperature of the reaction mixture exceeds 60 °

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1 bromo 4 nitrobenzene

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study with quizlet and memorize flashcards containing terms like reacting sulfuric and nitric acid together produces the nitronium ion alone nitric acid is not a strong enough electrophile to react with bromobenzene sulfuric acid is added to the

nitric acid the reverse rxn can overheat and boil easily includes a claisen adaptor and

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture hno 3 h 2 so 4 in the 125 ml flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 ml erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0 6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion no2 which acts as the electrophilic species

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lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with either a mixture of sulfuric acid or nitric acid creates what is known as a nitro group

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