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$$\begin{array}{l} \text{N}_2 + \text{F}_2 \rightarrow 2 \text{ NF}_3 \\ 2 \text{ C}_6\text{H}_{10} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O} \\ 3 \text{ HBr} + \text{KHCO}_3 \rightarrow \text{H}_2\text{O} + \text{KBr} + \text{CO}_2 \\ \text{GaBr}_3 + \text{NaBr} \rightarrow \text{SnO}_4 + \text{Na}_2\text{SO}_3 + \text{Ga}_2\text{SO}_3 + 5 \text{ NF}_3 + \text{SnF}_2 + \text{N}_2\text{O}_3 \end{array}$$
 using the equation from problem 2 above answer the following

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