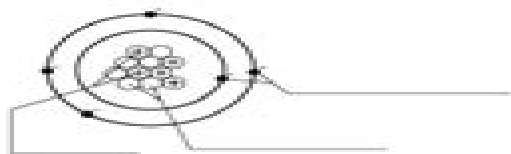


# Basic Atomic Structure

1. Label the parts of an atom in the diagram below



- What type of charge does a proton have? \_\_\_\_\_
- What type of charge does a neutron have? \_\_\_\_\_
- What type of charge does an electron have? \_\_\_\_\_
- Which two subatomic particles are located in the nucleus of an atom? \_\_\_\_\_

2. An element is represented by its chemical symbol along with a few numbers. Based on the example on the right, fill in the numbers of protons (P), neutrons (N), and electrons (E) for the following elements.

|        |                       |
|--------|-----------------------|
| 8      | ← Atomic Number       |
| O      | ← Symbol              |
| Oxygen | ← Name                |
| 15.999 | ← Average atomic mass |

|        |          |
|--------|----------|
| 6      | #P _____ |
| C      | #N _____ |
| Carbon | #E _____ |
| 12.011 |          |

|        |          |
|--------|----------|
| 10     | #P _____ |
| Ne     | #N _____ |
| Neon   | #E _____ |
| 20.108 |          |

|           |          |
|-----------|----------|
| 19        | #P _____ |
| K         | #N _____ |
| Potassium | #E _____ |
| 39.098    |          |

3. Complete the table using your knowledge of the periodic table.

| Symbol | Atomic Number | Mass Number | Number of Protons | Number of Electrons | Number of Neutrons |
|--------|---------------|-------------|-------------------|---------------------|--------------------|
| Na     |               |             | 11                |                     | 12                 |
| K      |               | 39          |                   | 19                  |                    |
|        |               |             | 38                |                     | 50                 |
| F      |               |             |                   | 9                   | 10                 |
|        | 20            | 40          |                   | 20                  |                    |
|        | 50            |             |                   | 50                  | 69                 |
| I      | 53            | 127         |                   |                     |                    |

# Atomic Structure Exercise 3 Answers

**Emil J. Margolis**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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