

Atomic Structure Worksheet

1. Name the three particles of the atom and their respective charges.

Particle	Charge type	Charge

2. The nucleus has a _____ charge and consist of _____ and _____.
3. Electrons are located in the _____ of an atom and carry _____ charge.
4. The number of protons in one atom of an element determines the atom's _____, and the number of electrons determines _____ of an element.
5. The atomic number tells you the number of _____ in one atom of an element. It also tells you the number of _____ in a neutral atom of that element. The atomic number gives the "identity" of an element as well as its location on the Periodic Table. No two different elements will have the _____ atomic number.
6. Give the symbol and number of protons in one atom of:
- | | | | |
|-----------------|----------------|-----------------|--------------|
| Beryllium _____ | Fluorine _____ | Magnesium _____ | Sulfur _____ |
| Cobalt _____ | Tin _____ | Arsenic _____ | Gold _____ |
7. Give the symbol and number of electrons in a neutral atom of:
- | | | |
|-----------------|---------------|---------------|
| Potassium _____ | Argon _____ | Calcium _____ |
| Chromium _____ | Cadmium _____ | Lead _____ |
8. Give the isotope symbol and number of neutrons in one atom of the following elements. Show your calculations.
- | | |
|-----------------------|------------------------|
| Cesium - 133 _____ | Iron - 54 _____ |
| Ruthenium - 106 _____ | Nickel - 58 _____ |
| Polonium - 209 _____ | Iodine - 135 _____ |
| Indium - 115 _____ | Phosphorous - 31 _____ |

Atomic Structure Answers Section 1

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