

## Key Concepts in Chemistry – Atomic Structure

1 – What was the main reason that the Dalton model of an atom changed over time?

- a) The discovery of subatomic particles
- b) The use of computer models
- c) Improvement in electron microscope technology
- d) The discovery of radioactivity

2 – Which of these best describes the structure of an atom?

- a) A small sphere containing protons, neutrons and electrons
- b) A small sphere with electrons on the outside and protons and neutrons inside it
- c) A central nucleus containing protons and neutrons surrounded by electrons in shells
- d) A central nucleus containing electrons and protons surrounded by neutrons in shells

3 – Which of these best describes the mass and charge of a proton?

- a) A mass of 2 and a charge of +1
- b) A mass of 1 and a charge of +1
- c) A mass of 0 and a charge of +1
- d) A mass of 1 and a charge of -1

4 – Which of these best describes the mass and charge of a neutron?

- a) A mass of 2 and a charge of +1
- b) A mass of 1 and a charge of +1
- c) A mass of 0 and a charge of +1
- d) A mass of 1 and a charge of 0

5 – Which of these best describes the mass and charge of an electron?

- a) A mass which is negligible and a charge of -1
- b) A mass of 1 and a charge of +1
- c) A mass of 0 and a charge of -1
- d) A mass which is negligible and a charge of +1

6 – Atoms have no overall charge because...

- a) They do not contain charged particles
- b) They contain neutrons which have no charge
- c) They contain equal numbers of positive protons and negative electrons
- d) They contain equal numbers of protons, neutrons and electrons

# Atomic Structure And Periodicity Multiple Choice Questions

**YCT Expert Team**



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