

# COVALENT BONDING

Name \_\_\_\_\_

Covalent bonding occurs when two or more nonmetals share electrons, attempting to attain a stable octet of electrons at least part of the time. For example:

H wants 2 electrons  
all other atoms 8



Note that hydrogen is content with 2, not 8, electrons

6 electrons  
in the middle  
share

Show how covalent bonding occurs in each of the following pairs of atoms. Atoms may share one, two or three pairs of electrons.

1. H + H (H<sub>2</sub>)  
W = 1 + 1 = 2  
A = 2  
S = 2

2. F + F (F<sub>2</sub>)  
W = 8 + 8 = 16  
A = 7 + 7 = 14  
S = 2

3. O + O (O<sub>2</sub>)  
W = 8 + 8 = 16  
A = 6 + 6 = 12  
S = 4

4. N + N (N<sub>2</sub>)  
W = 8 + 8 = 16  
A = 5 + 5 = 10  
S = 6

5. C + O (CO)  
W = 8 + 8 = 24  
A = 4 + 6 = 10  
S = 8

6. H + O (H<sub>2</sub>O)  
W = 2 + 2 + 8 = 12  
A = 1 + 1 + 6 = 8  
S = 4



## 6 2 Covalent Bonding Answer Key

**David Margolese,David W. Oxtoby**

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### **Table of Contents 6 2 Covalent Bonding Answer Key**

1. Understanding the eBook 6 2 Covalent Bonding Answer Key
  - The Rise of Digital Reading 6 2 Covalent Bonding Answer Key
  - Advantages of eBooks Over Traditional Books
2. Identifying 6 2 Covalent Bonding Answer Key
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an 6 2 Covalent Bonding Answer Key
  - User-Friendly Interface
4. Exploring eBook Recommendations from 6 2 Covalent Bonding Answer Key
  - Personalized Recommendations
  - 6 2 Covalent Bonding Answer Key User Reviews and Ratings
  - 6 2 Covalent Bonding Answer Key and Bestseller Lists
5. Accessing 6 2 Covalent Bonding Answer Key Free and Paid eBooks

- 6 2 Covalent Bonding Answer Key Public Domain eBooks
- 6 2 Covalent Bonding Answer Key eBook Subscription Services
- 6 2 Covalent Bonding Answer Key Budget-Friendly Options

6. Navigating 6 2 Covalent Bonding Answer Key eBook Formats

- ePUB, PDF, MOBI, and More
- 6 2 Covalent Bonding Answer Key Compatibility with Devices
- 6 2 Covalent Bonding Answer Key Enhanced eBook Features

7. Enhancing Your Reading Experience

- Adjustable Fonts and Text Sizes of 6 2 Covalent Bonding Answer Key
- Highlighting and Note-Taking 6 2 Covalent Bonding Answer Key
- Interactive Elements 6 2 Covalent Bonding Answer Key

8. Staying Engaged with 6 2 Covalent Bonding Answer Key

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers 6 2 Covalent Bonding Answer Key

9. Balancing eBooks and Physical Books 6 2 Covalent Bonding Answer Key

- Benefits of a Digital Library
- Creating a Diverse Reading Collection 6 2 Covalent Bonding Answer Key

10. Overcoming Reading Challenges

- Dealing with Digital Eye Strain
- Minimizing Distractions
- Managing Screen Time

11. Cultivating a Reading Routine 6 2 Covalent Bonding Answer Key

- Setting Reading Goals 6 2 Covalent Bonding Answer Key
- Carving Out Dedicated Reading Time

12. Sourcing Reliable Information of 6 2 Covalent Bonding Answer Key

- Fact-Checking eBook Content of 6 2 Covalent Bonding Answer Key
- Distinguishing Credible Sources

13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development

- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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