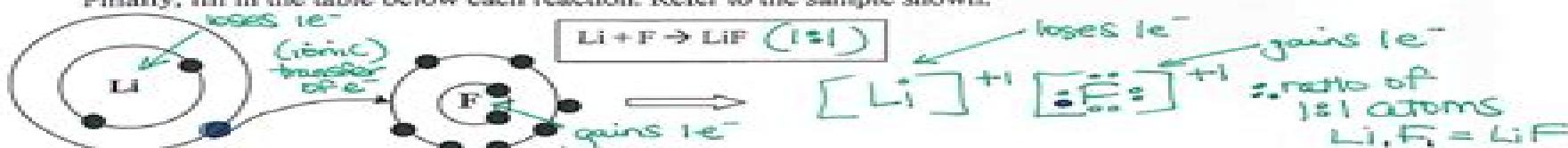


Ionic Bonding Worksheet

For each pair of elements below draw an atomic diagram showing electrons in different energy levels. Draw arrows to show where the outer electrons will go during a chemical reaction, then draw the resulting compound. Finally, fill in the table below each reaction. Refer to the sample shown.



Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
Li	1ve	transfers 1ve to F	Li ⁺
F	7ve	accepts 1ve from Li	F ⁻

Reactions	Atoms	Valence electrons	Electron transfer from/to each atom	Ions formed in the product
1) Li + Cl \rightarrow LiCl	Li	1ve	donates 1e ⁻	Li ⁺
	Cl	7ve	accepts 1e ⁻	Cl ⁻
2) Ca + O \rightarrow CaO	Ca	2ve	donates 2e ⁻	Ca ⁺
	O	6ve	accepts 2e ⁻	O ²⁻
3) Be + F \rightarrow BeF ₂	Be	2ve	donates 2e ⁻ to each F.	Be ²⁺
	F	7ve	each F accepts 1ve each	F ⁻
4) Mg + S \rightarrow MgS	Mg	2ve	donates 2e ⁻ to S	Mg ²⁺
	S	6ve	accepts 2e ⁻	S ²⁻
5) K + F \rightarrow KF	K	1ve	donates 1e ⁻	K ⁺
	F	7ve	accepts 1e ⁻	F ⁻

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Table of Contents 6 1 Ionic Bonding Worksheet Answers

1. Understanding the eBook 6 1 Ionic Bonding Worksheet Answers
 - The Rise of Digital Reading 6 1 Ionic Bonding Worksheet Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying 6 1 Ionic Bonding Worksheet Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 6 1 Ionic Bonding Worksheet Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from 6 1 Ionic Bonding Worksheet Answers
 - Personalized Recommendations
 - 6 1 Ionic Bonding Worksheet Answers User Reviews and Ratings
 - 6 1 Ionic Bonding Worksheet Answers and Bestseller Lists
5. Accessing 6 1 Ionic Bonding Worksheet Answers Free and Paid eBooks
 - 6 1 Ionic Bonding Worksheet Answers Public Domain eBooks
 - 6 1 Ionic Bonding Worksheet Answers eBook Subscription Services
 - 6 1 Ionic Bonding Worksheet Answers Budget-Friendly Options

6. Navigating 6 1 Ionic Bonding Worksheet Answers eBook Formats
 - ePUB, PDF, MOBI, and More
 - 6 1 Ionic Bonding Worksheet Answers Compatibility with Devices
 - 6 1 Ionic Bonding Worksheet Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 6 1 Ionic Bonding Worksheet Answers
 - Highlighting and Note-Taking 6 1 Ionic Bonding Worksheet Answers
 - Interactive Elements 6 1 Ionic Bonding Worksheet Answers
8. Staying Engaged with 6 1 Ionic Bonding Worksheet Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 6 1 Ionic Bonding Worksheet Answers
9. Balancing eBooks and Physical Books 6 1 Ionic Bonding Worksheet Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 6 1 Ionic Bonding Worksheet Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine 6 1 Ionic Bonding Worksheet Answers
 - Setting Reading Goals 6 1 Ionic Bonding Worksheet Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of 6 1 Ionic Bonding Worksheet Answers
 - Fact-Checking eBook Content of 6 1 Ionic Bonding Worksheet Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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