

- 6 (a) i. Using chemical equations only, distinguish between ethene and ethyne. (3 marks)
- ii. Name the compound formed when ethyne polymerises. (1 mark)
- iii. Write the structure of the compound in a(ii). (1 mark)
- (b) i. What will be observed when:
- (I) Fehling's solution is added to reducing sugar?
 - (II) Millon's reagent is added to protein?
 - (III) Sodium metal is dropped into a beaker containing ethanol?
- (3 marks)
- (ii) Calculate the solubility in mol/dm^3 of 4.9 g KClO_3 in 20 cm^3 of water at 66°C .
 $[\text{K} = 39, \text{Cl} = 35.5, \text{O} = 16]$ (5 marks)
- (c) i. State **three** physical properties of ethanol. (3 marks)
- ii. Write the equations for the preparation of ethanol from ethene. (4 marks)
- (d) i. Mention **two** uses of sodium trioxocarbonate (IV). (2 marks)
- ii. State **three** properties of transition elements. (3 marks)

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Albert A Gayle



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