

6 (a)i. Using chemical equations only, distinguish between ethene and ethyne. (3 marks)

ii. Name the compound formed when ethyne polymerises. (1 mark)

iii. Write the structure of the compound in a(ii). (1 mark)

(b)i. What will be observed when:

(I) Fehling's solution is added to reducing sugar?
(II) Millon's reagent is added to protein?
(III) Sodium metal is dropped into a beaker containing ethanol? (3 marks)

(ii) Calculate the solubility in mol/dm³ of 4.9 g KClO₃ in 20 cm³ of water at 66°C.
[K = 39, Cl = 35.5, O = 16] (5 marks)

(c)i. State three physical properties of ethanol. (3 marks)

ii. Write the equations for the preparation of ethanol from ethene. (4 marks)

(d)i. Mention two uses of sodium trioxocarbonate (IV). (2 marks)

ii. State three properties of transition elements. (3 marks)

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Albert A Gayle

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