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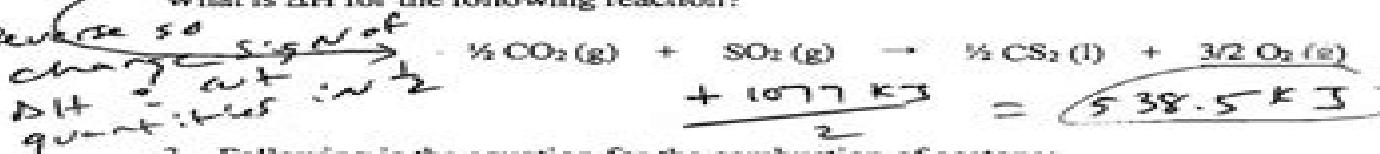
1. Hydrogen,  $H_2$ , is used as rocket fuel. The hydrogen is burned in oxygen,  $O_2$ , according to the following equation. What is the enthalpy change if 200.0 g of hydrogen,  $H_2$  is burned?



2. Carbon disulfide burns in air, producing carbon dioxide and sulfur dioxide.



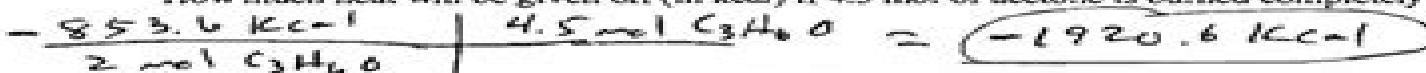
What is  $\Delta H$  for the following reaction?



3. Following is the equation for the combustion of acetone:



How much heat will be given off (in kcal) if 4.5 mol of acetone is burned completely?



4. The specific heat of iron is 0.110 cal / g  $^{\circ}\text{C}$ . How much heat must be added to 5.2 g of iron at 25  $^{\circ}\text{C}$  to raise its temperature to 225  $^{\circ}\text{C}$ ?



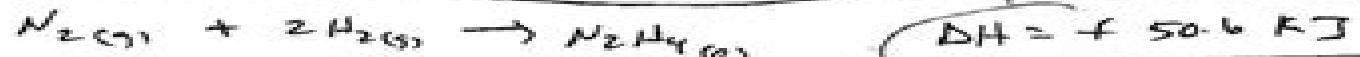
5. Hydrazine,  $N_2H_4$ , is a colorless liquid used as a rocket fuel. What is the enthalpy change for the process in which hydrazine is formed from its elements (reaction (1) below)? Use Hess's Law to calculate  $\Delta H$  for reaction (1) given the data below.



Reverse



double  
quantity



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**Henry Dorin**

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**Chemistry: Matter & Change, Solving Problems - A Chemistry Handbook** McGraw Hill,2001-08 Glencoe Chemistry Solving Problems A Chemistry Handbook

Matter and Change *Ebook: Chemistry* Julia Burdge,2014-10-16 Chemistry Third Edition by Julia Burdge offers a clear writing style written with the students in mind Julia uses her background of teaching hundreds of general chemistry students per year and creates content to offer more detailed explanation on areas where she knows they have problems With outstanding art a consistent problem solving approach interesting applications woven throughout the chapters and a wide range of end of chapter problems this is a great third edition text

*Study Guide with Selected Solutions* Karen Timberlake,2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems

**Problem Solving Guide and Workbook for Introductory Chemistry by Steve Russo, Mike Silver** Saundra Yancy McGuire,Steve Russo,Mike Silver,2002 Provides over 175 worked examples and more than 500

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**Merrill Chemistry** Robert C. Smoot,Smoot,Richard G. Smith,Jack Price,1998 *Partial Solutions Manual* Darrell D. Ebbing,David Bookin,1998

*Living Chemistry* David A. Ucko,1986 **Student Solutions Manual** Darrell Ebbing,2005-03-17 Provides solutions to odd numbered Practice Problems General Problems and Cumulative Skills Problems plus answers to Review Questions

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