



Name _____ Class _____ Date _____

1

How many atoms of oxygen (O) are in the **chemical formula** below?

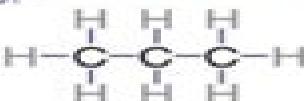
A 4
B 5
C 7
D 12



2

The diagram shown below is a **structural formula** for the gas propane. What would be the correct **chemical formula** for propane?

A C_2H_6
B CH_3
C $8\text{H}_3\text{C}$
D CH



3

Carbon (C) and hydrogen (H) create **covalent bonds** to share **electrons** between them. What is the total number of electrons being shared between **carbon and hydrogen** in the molecule of propane shown below?

A 8
B 11
C 16
D 32



4

The **reason** why the bonds between carbon (C) and hydrogen (H) are **covalent** is _____



A C and H do not have electrons to give away
B C and H are both nonmetals
C C and H are both metalloids
D C and H can bond only by sharing electrons

5

The diagram below represents the **structural formula** for several **water molecules**. How many **molecules** are represented?

A 3
B 5
C 10
D 15



6

Explain the **arrangement** of the molecules in the diagram of water below.

A it is a lucky arrangement
B water molecules are sticky
C water molecules have polarity
D water molecules are magnetic



7

Water molecules have some **polarity** in their structure. Using the diagram below, notice how the ends of H atoms (**positive**) and O atoms (**negative**) meet. This property is the **reason** that _____

A water forms droplets easily
B water boils easily
C ice melts easily
D water has color



8

Since H atoms are slightly **positive** and the O atoms are slightly **negative**, molecules of water have **polarity**. This makes water a **good** _____

A solute
B refrigerant
C solvent
D magnet



9

The chemical formula for the **oxygen** that we breathe is _____

A O
B O_2
C O_3
D 20



10

How many **molecules** of oxygen are represented in the chemical formula below?

A 2
B 4
C 6
D 18



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