

20.1**DESCRIBING ACIDS AND BASES**
SECTION REVIEW**Objectives**

- Use the properties of acids and bases
- Name an acid or base when given the formula

Key Terms

- acid
- base

Part A: Completion

After this completion exercise, you should know an alternative naming of acids and bases and how they are represented in the periodic table. Which block uses the name "perchloric" which is correct, above gallium, or below bismuth?

An aqueous solution of 1. has acidic properties. Acids contain 2. atoms in 3. the products are 4. and 5. are composed of a metal cation with one or more 6. and a 7.

Acids produced in 8. have when dissolved in water. There is a standard equation for naming acids. For example, HCl is called 9. acid, while HNO₃ is called 10. acid.

Basic properties 11. are produced when they dissolve in water. Bases correspond to these are 12. and named for the anions they are anion for example 13.

14.

Part B: True-False

Correctly decide whether statements are true or false. If true, write T; if false, write F.

15. Acids are protonic acids, containing hydrogen atoms in addition to metal cations.

16. Acids are acids because they contain hydroxide ions.

17. Chlorine forms a covalent hydride when reacting with a base.

20 Describing Acids And Bases Answer Key

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Table of Contents 20 Describing Acids And Bases Answer Key

1. Understanding the eBook 20 Describing Acids And Bases Answer Key
 - The Rise of Digital Reading 20 Describing Acids And Bases Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying 20 Describing Acids And Bases Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an 20 Describing Acids And Bases Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from 20 Describing Acids And Bases Answer Key
 - Personalized Recommendations
 - 20 Describing Acids And Bases Answer Key User Reviews and Ratings

- 20 Describing Acids And Bases Answer Key and Bestseller Lists
- 5. Accessing 20 Describing Acids And Bases Answer Key Free and Paid eBooks
 - 20 Describing Acids And Bases Answer Key Public Domain eBooks
 - 20 Describing Acids And Bases Answer Key eBook Subscription Services
 - 20 Describing Acids And Bases Answer Key Budget-Friendly Options
- 6. Navigating 20 Describing Acids And Bases Answer Key eBook Formats
 - ePUB, PDF, MOBI, and More
 - 20 Describing Acids And Bases Answer Key Compatibility with Devices
 - 20 Describing Acids And Bases Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of 20 Describing Acids And Bases Answer Key
 - Highlighting and Note-Taking 20 Describing Acids And Bases Answer Key
 - Interactive Elements 20 Describing Acids And Bases Answer Key
- 8. Staying Engaged with 20 Describing Acids And Bases Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers 20 Describing Acids And Bases Answer Key
- 9. Balancing eBooks and Physical Books 20 Describing Acids And Bases Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection 20 Describing Acids And Bases Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine 20 Describing Acids And Bases Answer Key
 - Setting Reading Goals 20 Describing Acids And Bases Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of 20 Describing Acids And Bases Answer Key
 - Fact-Checking eBook Content of 20 Describing Acids And Bases Answer Key
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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