

d. Ternary acids - just like binary acids, ternary acids must always contain hydrogen. These acids also contain a polyatomic ion.

Polyatomic ion	Name of ion	Polyatomic ion	Name of ion
OH^-	hydroxide	PO_4^{3-}	phosphate
O_2H^-	peroxide	CO_3^{2-}	carbonate
$\text{C}_2\text{H}_5\text{O}_2^-$	ethoxide	CH_3CO_2^-	acetate
SO_3^{2-}	sulfite	CH_3COO^-	acetate
SO_4^{2-}	sulfate	ClO_4^-	chlorate

Ternary acids are three element acids. Most ternary acids are **oxyacids** containing hydrogen, oxygen and other atoms.

Practice

- >Name the common acids (10 acids)
- Change the common ending to one of the following

Acid ending	10 acids replaced with
-ic	-ous acid
-ous	-ic acid

Formula	composition	Ternary compound name	Ternary acid name
H_2SO_4	SO_4^{2-}	hydrogen sulfate	Sulfuric acid
H_2SO_3	SO_4^{2-}	hydrogen sulfite	Sulfurous acid
H_2CO_3	CO_3^{2-}	hydrogen carbonate	Carbonic acid
H_2CO_4	CO_3^{2-}	hydrogen bicarbonate	Bicarbonate acid
H_2NO_2	NO_2^-	nitrogen dioxide	Nitric acid
H_2NO_3	NO_2^-	nitrogen trioxide	Nitrous acid
H_2NO_4	NO_2^-	nitrogen tetroxide	Nitric acid
H_2NO_5	NO_2^-	nitrogen pentoxide	Nitric acid

Answers For Writing Ternary Formulas

Charles H. Corwin

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