

## Balancing Chemical Reactions

Balance all of the chemical reactions below. All compounds are written correctly.

After balancing the reaction, determine the type(s) of chemical reaction(s).

TYPE(S) of RXN

1. 2  $\text{NH}_3 \rightarrow$  \_\_\_\_\_  $\text{N}_2 +$  3  $\text{H}_2$  Decomp
2. 4  $\text{Al} +$  3  $\text{O}_2 \rightarrow$  2  $\text{Al}_2\text{O}_3$  Syn/Comb
3. \_\_\_\_\_  $\text{N}_2\text{H}_4 +$  \_\_\_\_\_  $\text{O}_2 \rightarrow$  \_\_\_\_\_  $\text{N}_2 +$  2  $\text{H}_2\text{O}$  S.R. (?)
4. 2  $\text{Al} +$  3  $\text{H}_2\text{SO}_4 \rightarrow$  3  $\text{H}_2 +$  \_\_\_\_\_  $\text{Al}_2(\text{SO}_4)_3$  S.R.
5. \_\_\_\_\_  $\text{P}_2\text{O}_5 +$  3  $\text{H}_2\text{O} \rightarrow$  2  $\text{H}_3\text{PO}_4$  Syn
6. 1  $\text{C}_2\text{H}_2 +$  5  $\text{O}_2 \rightarrow$  4  $\text{CO}_2 +$  2  $\text{H}_2\text{O}$  Comb.
7. \_\_\_\_\_  $\text{Fe}_2\text{O}_3 +$  3  $\text{H}_2 \rightarrow$  2  $\text{Fe} +$  3  $\text{H}_2\text{O}$  S.R.
8. 4  $\text{FeS} +$  7  $\text{O}_2 \rightarrow$  2  $\text{Fe}_2\text{O}_3 +$  4  $\text{SO}_2$  Comb.
9. 1  $\text{Cu}_2\text{S} +$  3  $\text{O}_2 \rightarrow$  2  $\text{Cu}_2\text{O} +$  2  $\text{SO}_2$  Comb.
10. 4  $\text{NH}_3 +$  5  $\text{O}_2 \rightarrow$  4  $\text{NO} +$  6  $\text{H}_2\text{O}$  Comb.
11. \_\_\_\_\_  $\text{C}_3\text{H}_8 +$  5  $\text{O}_2 \rightarrow$  3  $\text{CO}_2 +$  4  $\text{H}_2\text{O}$  Comb
12. 2  $\text{AlPO}_4 +$  3  $\text{Na}_2\text{S} \rightarrow$  \_\_\_\_\_  $\text{Al}_2\text{S}_3 +$  2  $\text{Na}_3\text{PO}_4$  D.R.
13. \_\_\_\_\_  $\text{C}_{25}\text{H}_{52} +$  38  $\text{O}_2 \rightarrow$  26  $\text{H}_2\text{O} +$  25  $\text{CO}_2$  Comb
14. 2  $\text{FeCl}_3 +$  3  $\text{Na}_2\text{CO}_3 \rightarrow$  \_\_\_\_\_  $\text{Fe}_2(\text{CO}_3)_3 +$  6  $\text{NaCl}$  D.R.
15. \_\_\_\_\_  $\text{P}_4\text{O}_{10} +$  6  $\text{H}_2\text{O} \rightarrow$  4  $\text{H}_3\text{PO}_4$  Syn.
16. 4  $\text{NH}_3 +$  7  $\text{O}_2 \rightarrow$  4  $\text{NO}_2 +$  6  $\text{H}_2\text{O}$  Comb.
17. 2  $\text{C}_7\text{H}_8\text{O}_2 +$  15  $\text{O}_2 \rightarrow$  14  $\text{CO}_2 +$  6  $\text{H}_2\text{O}$  Comb.

# Answers For Stars Suite Chemistry

**Justin P. Lomont, Ian C. Stewart**



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